



PETRUS- ANNETTE PhD and Early-Stage Researchers Conference 2017

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What are PETRUS and
ANNETTE?

About the PETRUS-
ANNETTE 2017
conference

◆ Programme and
presentations

Registration Form

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Venue

Programme and presentations

Monday 26

14:00 - 18:00

- **Welcome**
Representatives from PETRUS/ANNETTE organizing committee
- **Lecture of Prof. F.J. Elorza (UPM)**
Hydro-mechanical and transport processes
- **Lecture of Dr. I. Paiva and Dr. M.J. Reis (IST)**
Radiation protection issues
- **Lecture of Prof. B. Bazargan Sabet (UL)**
Front-end nuclear fuel cycle waste

Tuesday 27

9:00 - 18:30

- **PhD/early stage researcher presentations**

-----LUNCH-----

- **Lecture of Prof. A. Gens (UPC)**
Coupled THM analysis for radioactive waste geological disposal
- **Lecture of Prof. J.M. Montel (UL)**
Natural analogue studies in the geological disposal of radioactive wastes
- **Lecture of Dr. Erika Neeft (COVRA)**
Carbon-14 Source Term
- **Lecture of Prof. K. Pedersen (Micans)**
Bacterial life in clay barriers surrounding radioactive waste in geological repositories

POWERED BY *strikingly*

CArbon-14 Source Term CAST

Name: Erika Neeft

Organisation: COVRA (WMO)

Date: 27 June 2017



The project has received funding from the European Union's Seventh Framework Programme for research, technological development and demonstration under grant agreement no. 604779, the CAST project.



Content

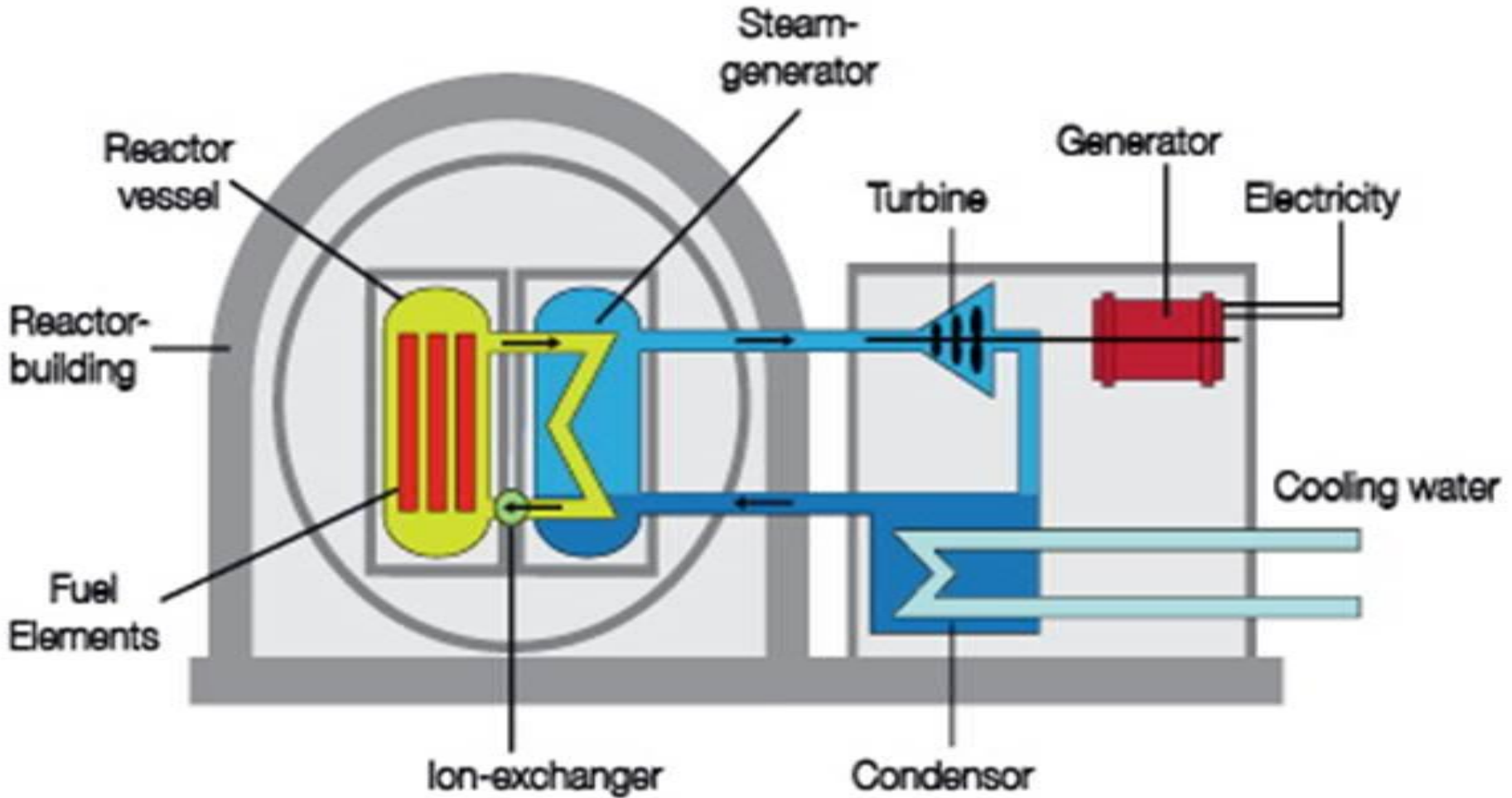


- Radionuclides
 - Generation in a nuclear plant
 - Fission, activation
 - Left for (geological) disposal
 - ETM and DTM
 - carbon-14
 - » Clearance level carbon-14 in waste in EU
- Disposal of waste
 - Potential migration of released radionuclides
 - Gas, dissolved, retarded
 - Natural carbon-14
 - Generation
 - Exposure
 - Potential exposure mechanism artificial carbon-14 if released as gas
 - Carbon-14 Source Term
 - Types of waste investigated
 - Potential release mechanisms at (geological) disposal conditions
 - Cementitious materials

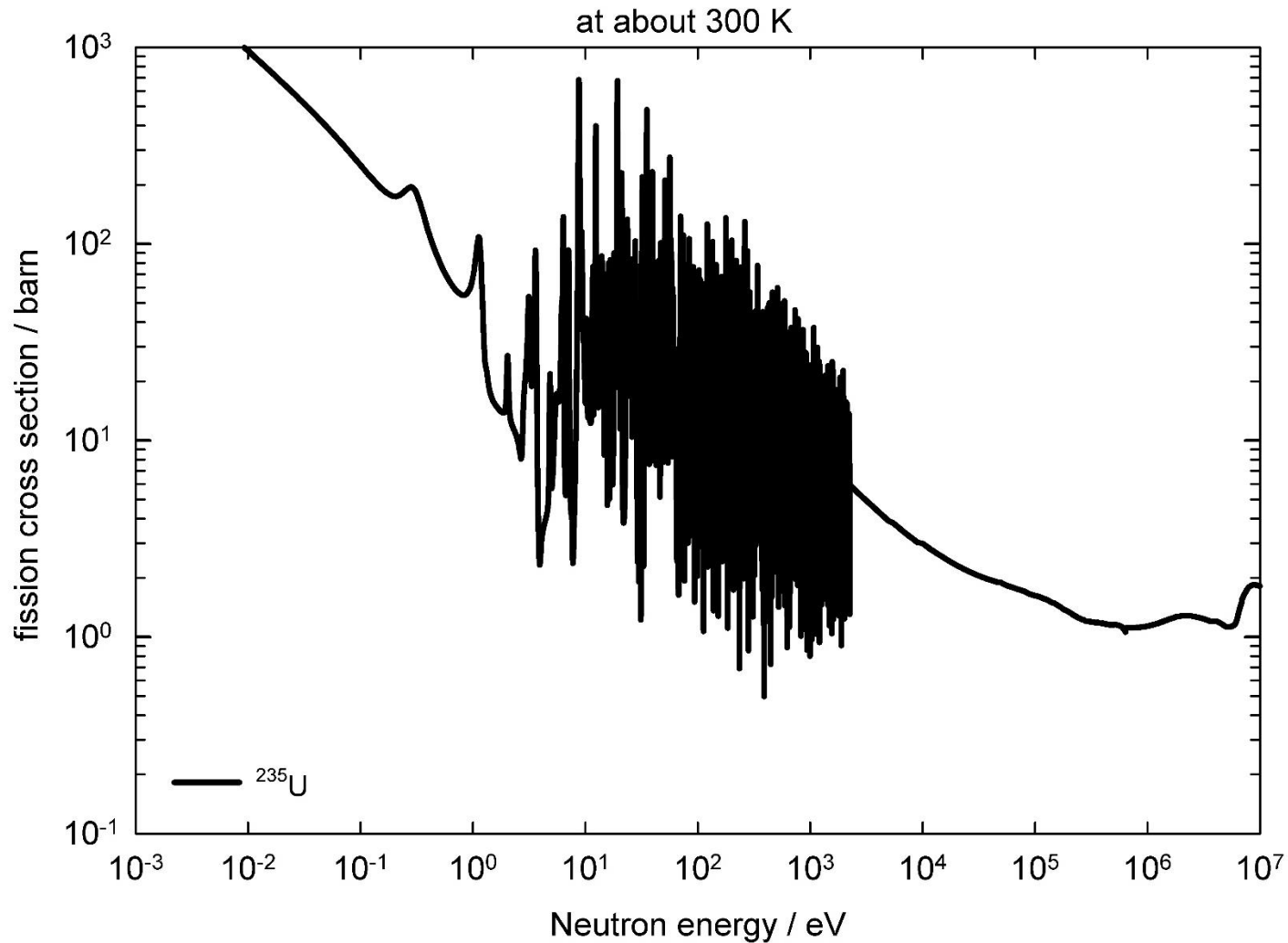
Generation



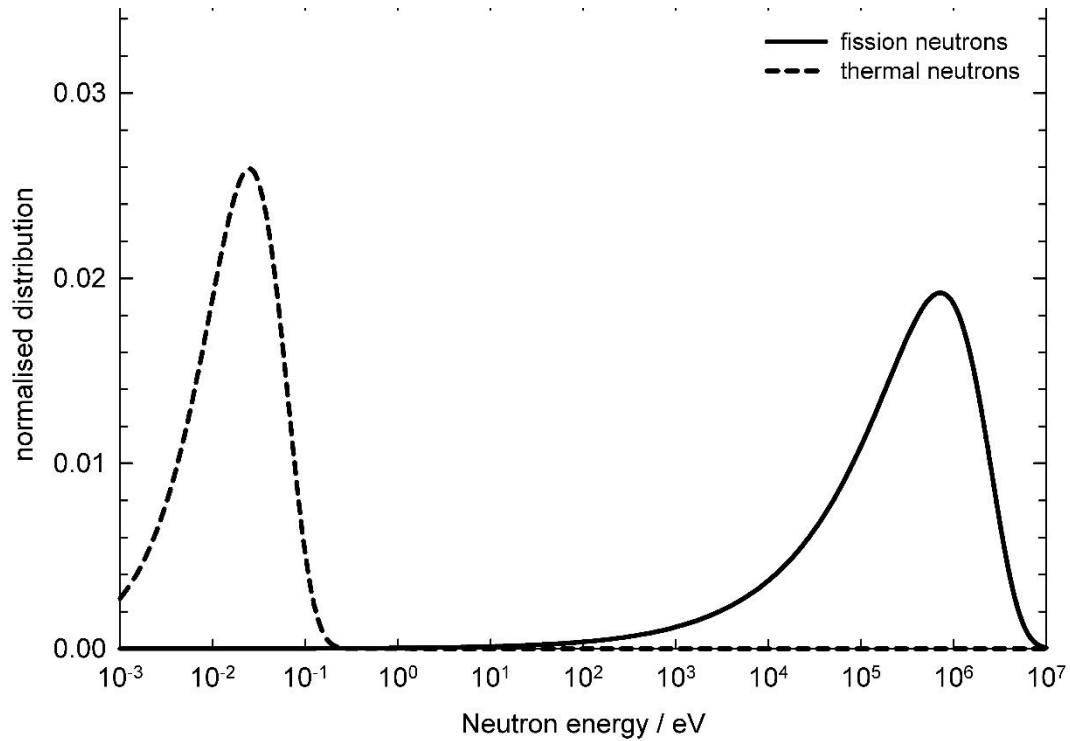
PWR



Generation

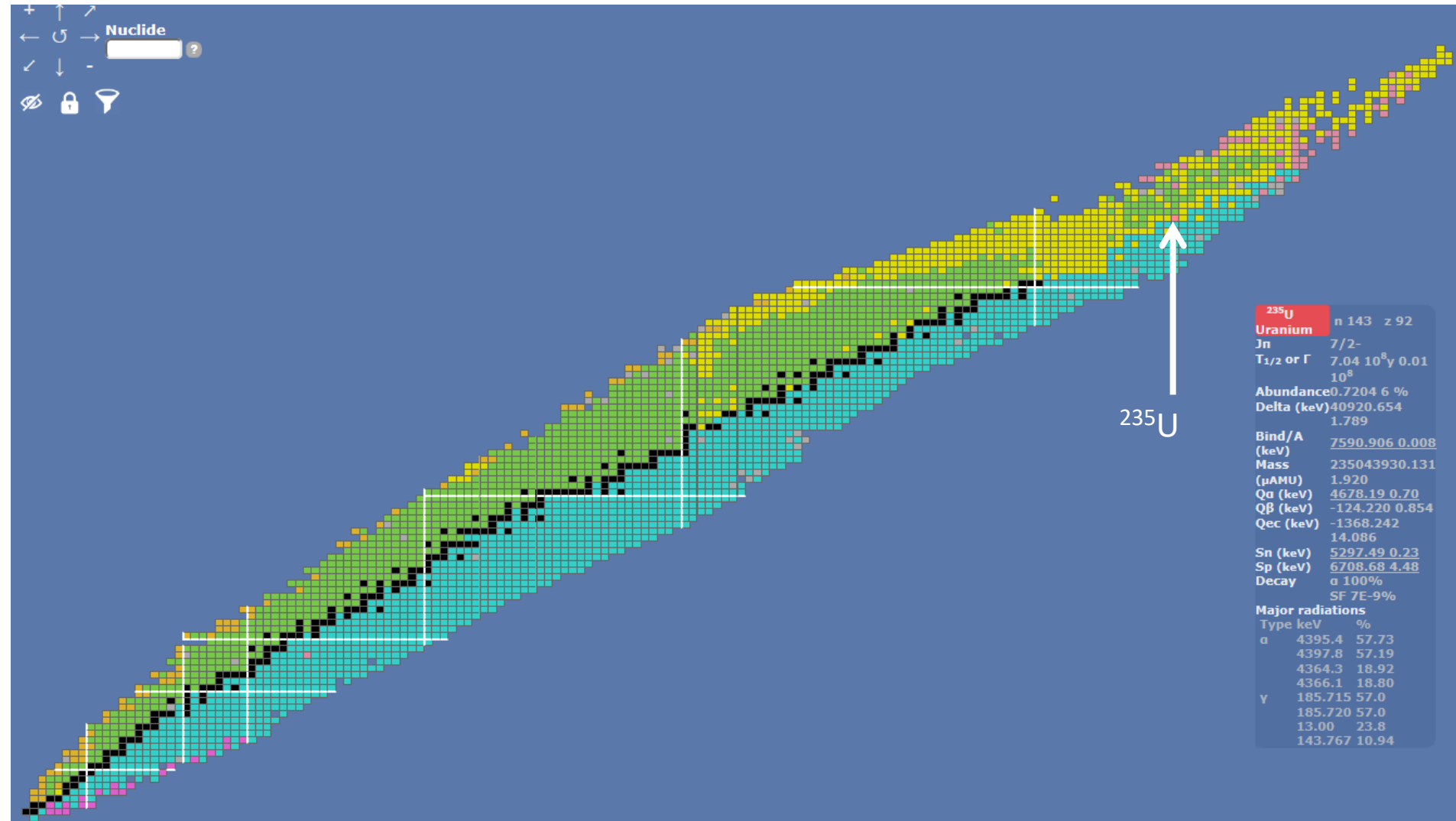


Generation



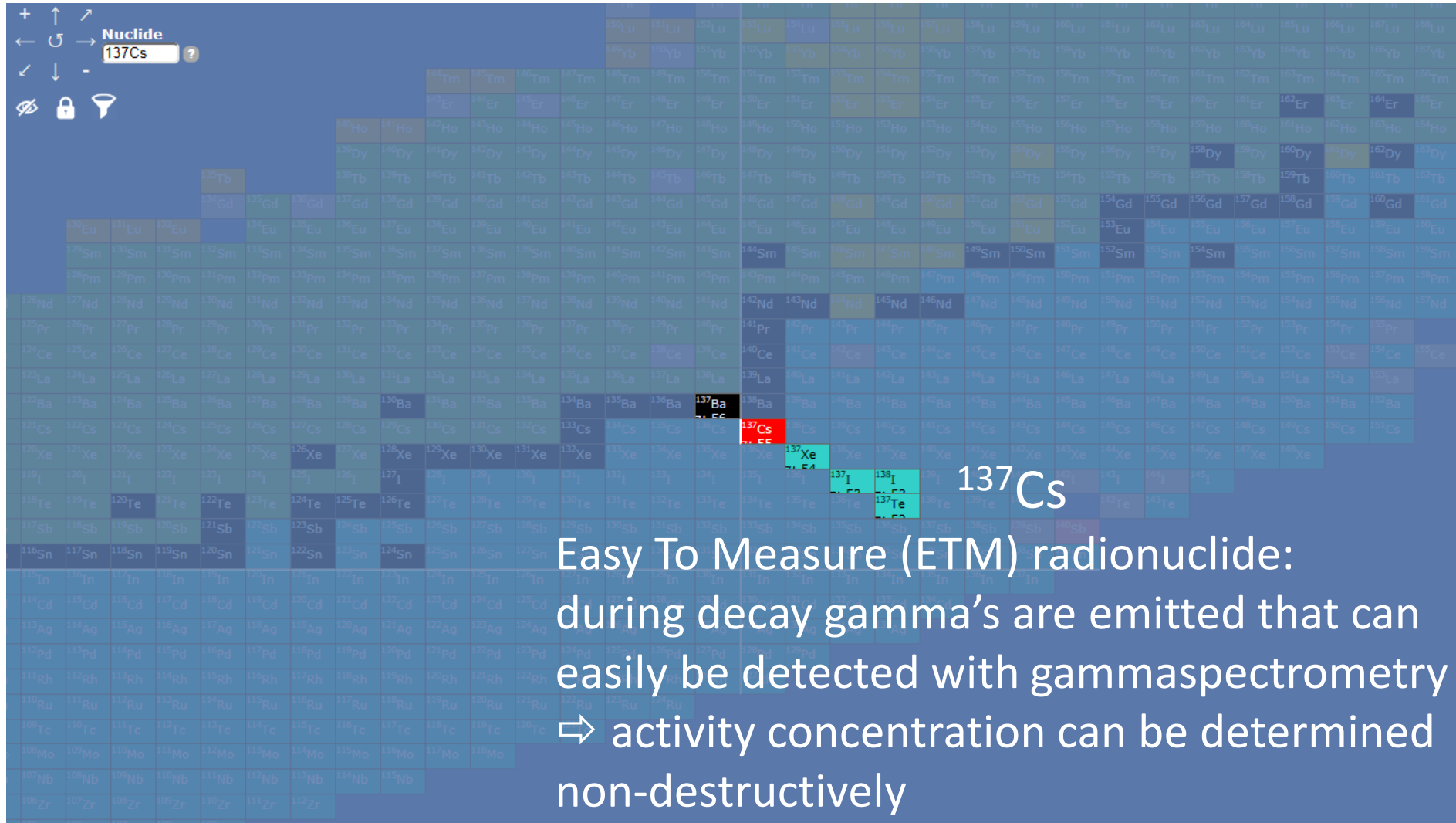


Generation

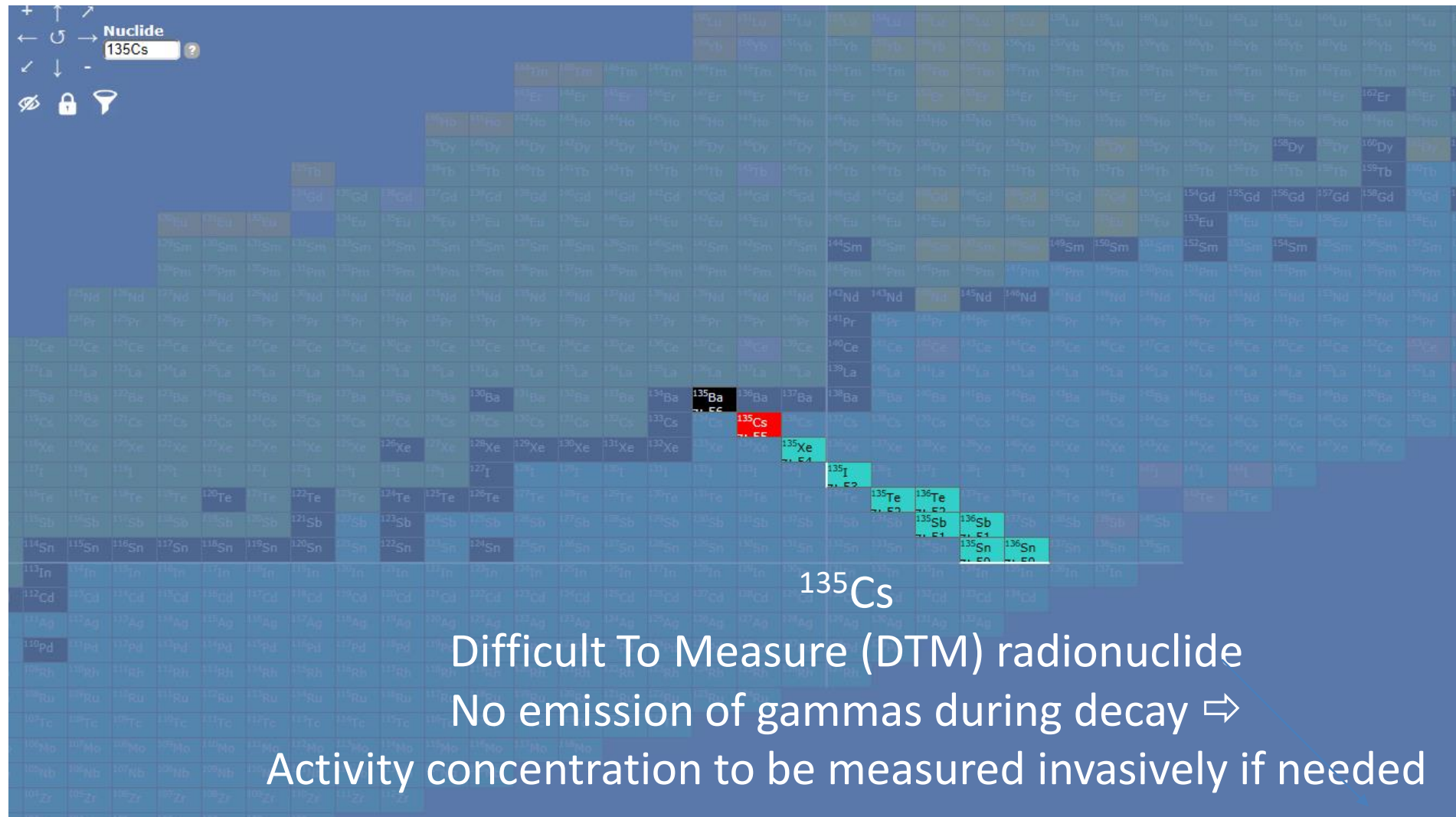


Live chart from IAEA, free online

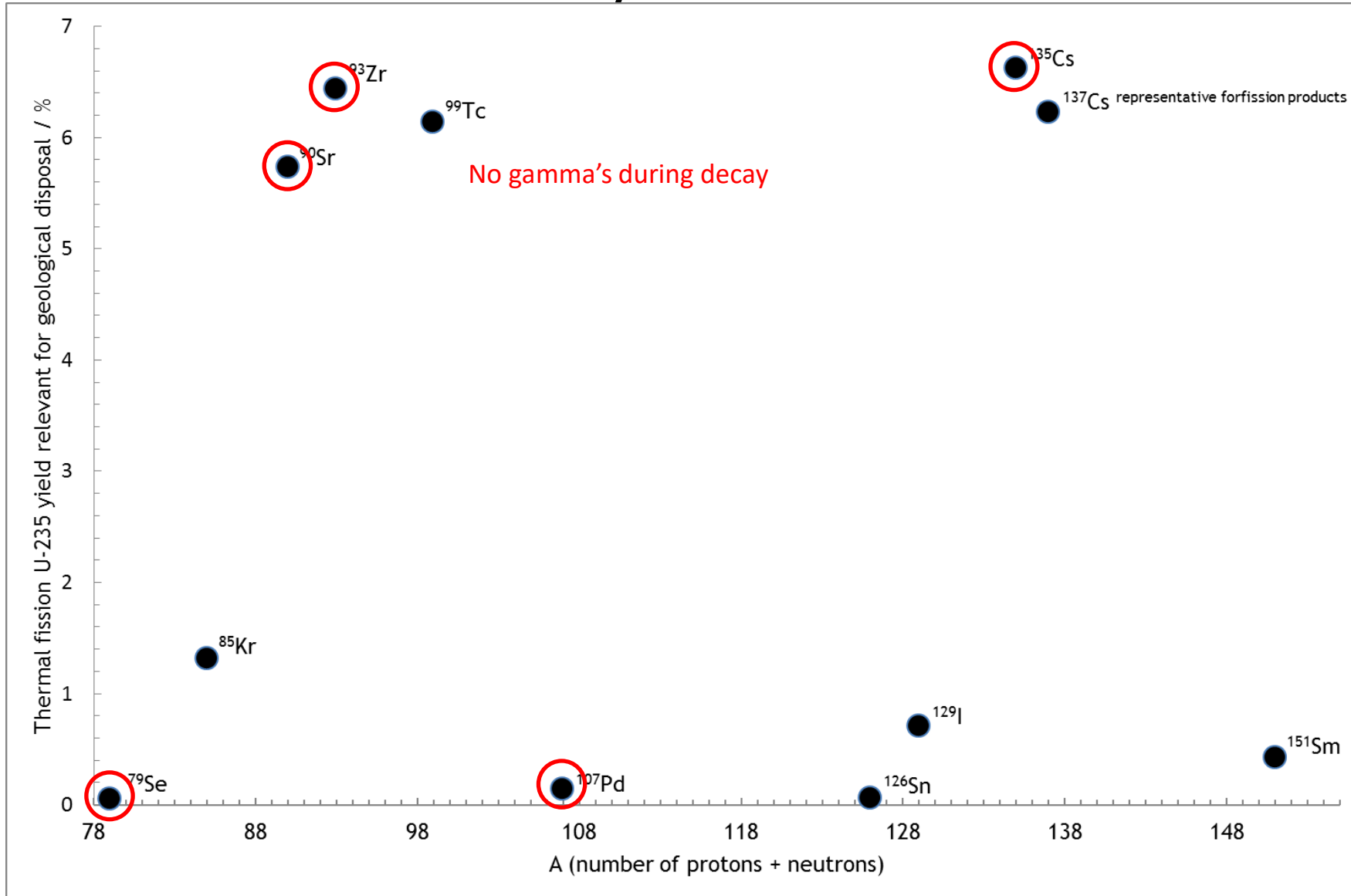
Example RN left for disposal from decay and fission



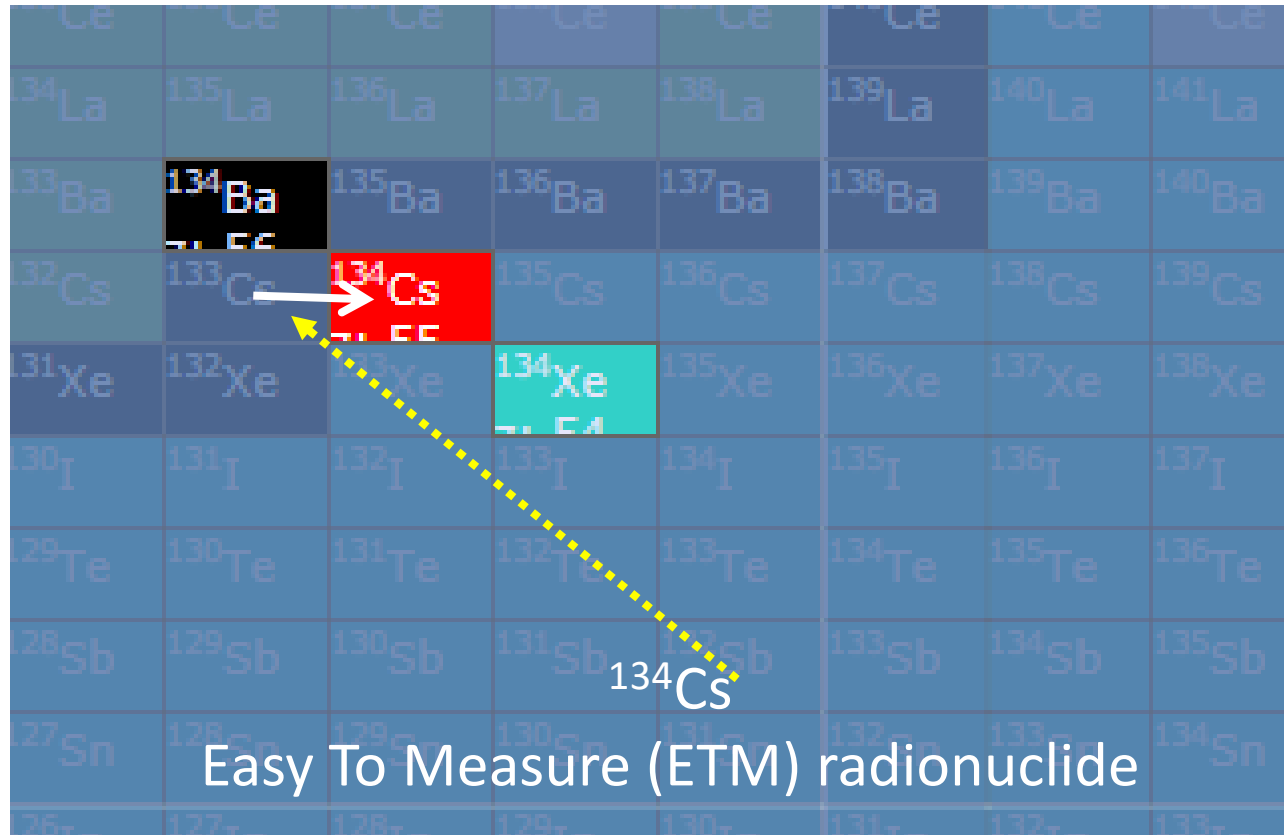
Example RN left for disposal from decay and fission



Examples RN left for disposal from decay and fission

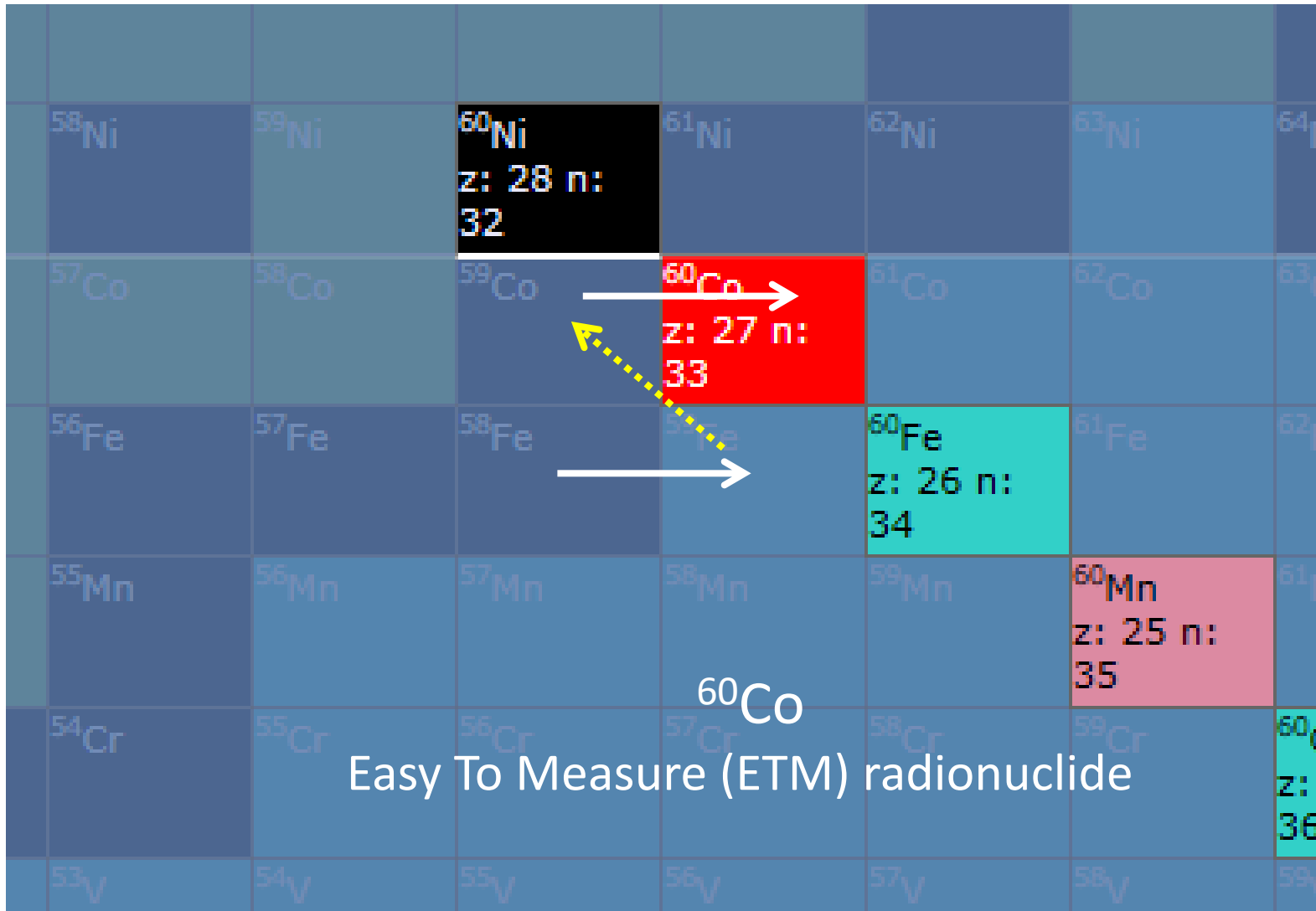


Example RN left from fission, decay and activation





Example RN left perhaps for disposal from activation





COVRA's storage period at least 100 years:
Fraction in activity left: $\left\{\frac{1}{2}\right\}^{100/t_{0,5}}$ for $^{60}\text{Co}=0,0000019$
i.e. reduction of a million



Neutron activation



- Identification activation path to obtain the precursors
- Knowledge of the chemical content of precursors
 - Can be impurities



Is RN relevant for disposal?

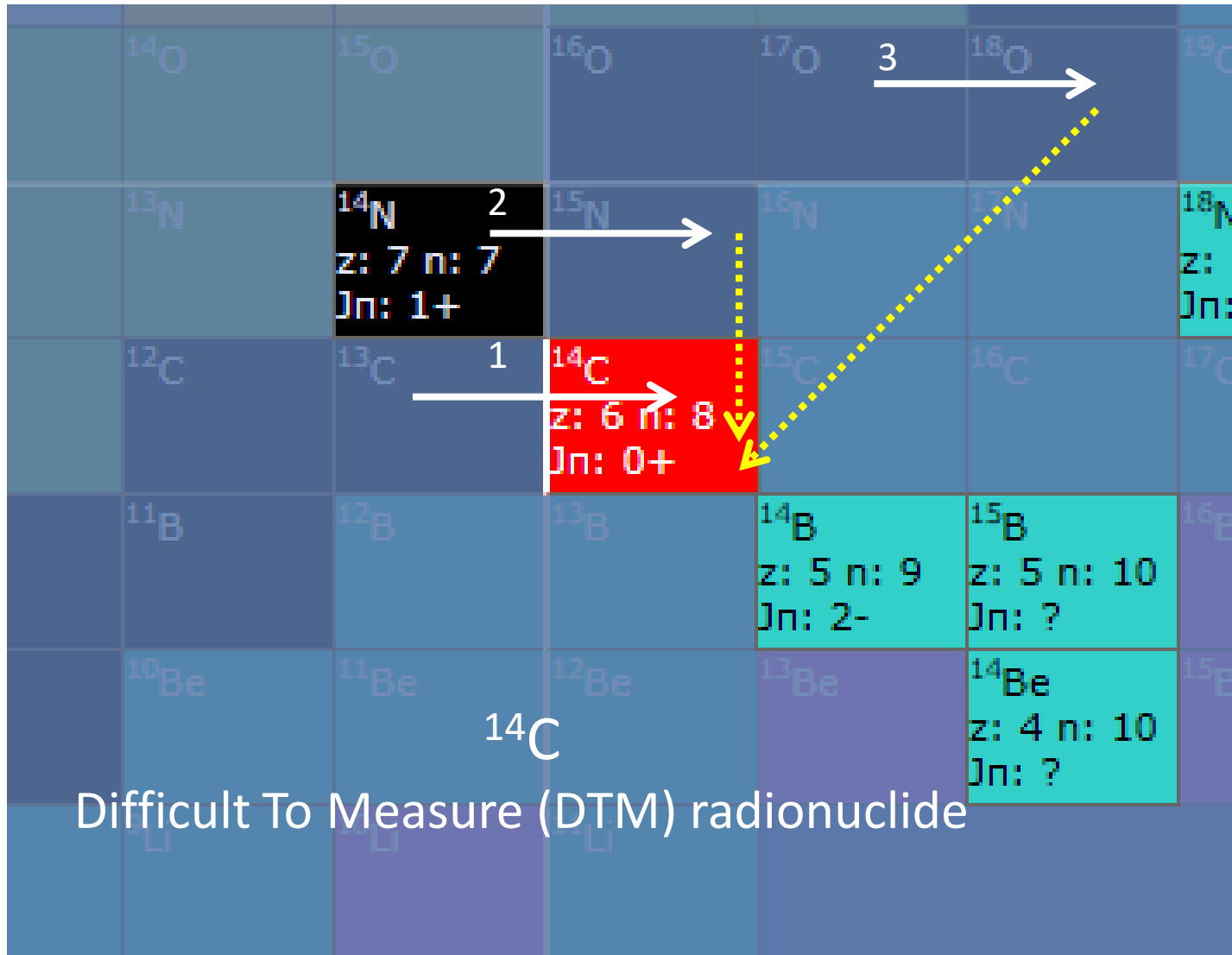
- Clearance levels in EU:
 - Council Directive 2013/59/EURATOM of 5 December 2013 laying down basic safety standards for protection against the dangers arising from exposure to ionising radiation, and repealing Directives 89/618/Euratom, 90/641/Euratom, 96/29/Euratom, 97/43/Euratom and 2003/122/Euratom, Official Journal of the European Union, L13/1-73, 17.1.2014
 - ^{14}C : 1 Bq per gram solid matter for example
0.000024 ppm in iron



COVRA's storage period at least 100 years:
Fraction in activity left: $\{1/2\}^{100/t_{0,5}}$ for $^{14}\text{C}=0,99$
i.e. no significant reduction after this storage period

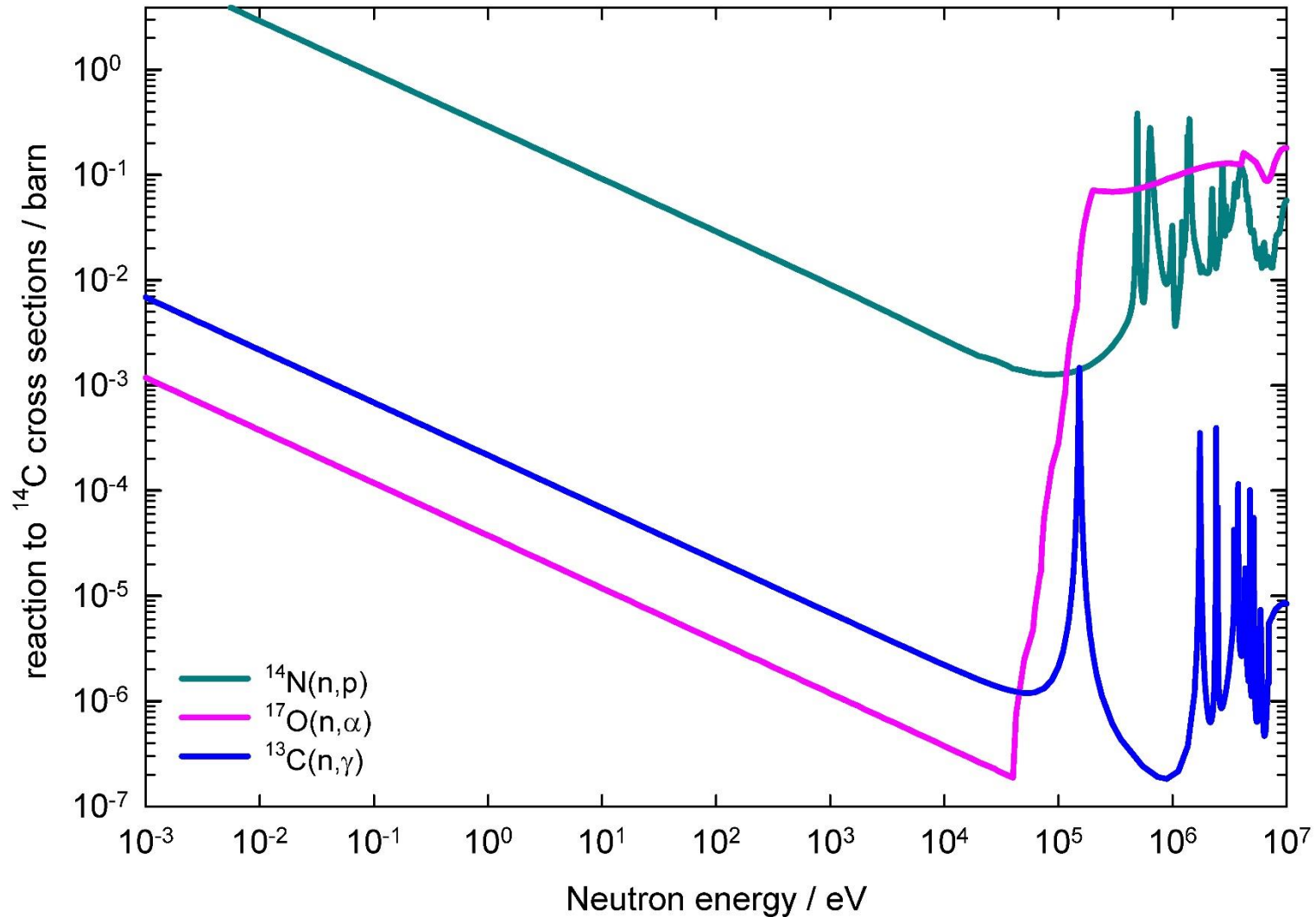


Neutron activation



Neutron activation

at about 300 K





Neutron activation

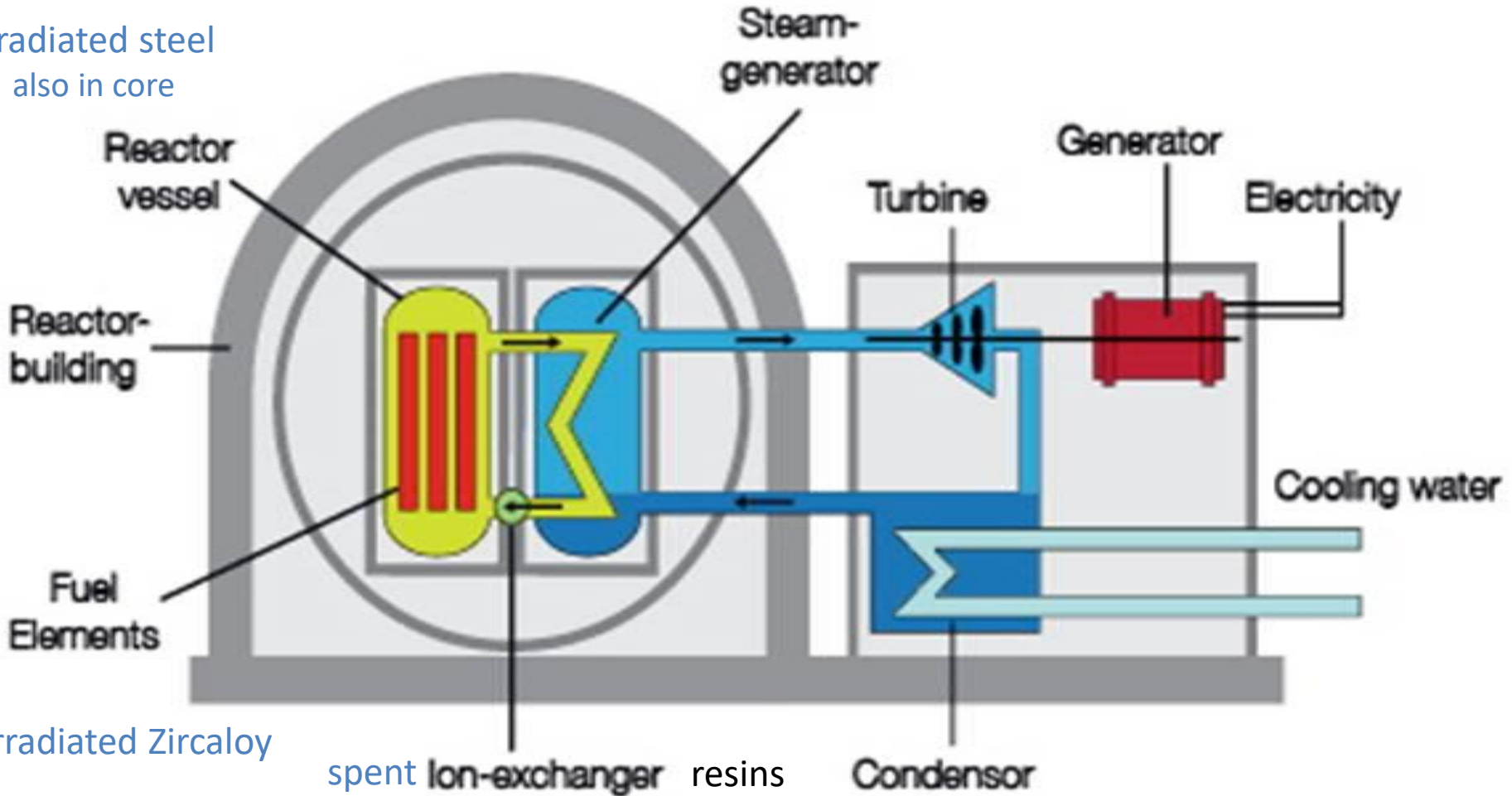


- Natural abundances
 - Nitrogen-14 : 99.636%
 - Oxygen-17 : 0.038%
 - Carbon-13 : 1.07%
- Natural abundance + thermal cross sections for the same carbon-14 contribution:
 - Chemical content carbon $\gg 10^5$ chemical nitrogen content
 - Chemical content oxygen $\gg 10^7$ chemical nitrogen content

Types of waste investigated



Irradiated steel
also in core





Carbon-14 act.conc.

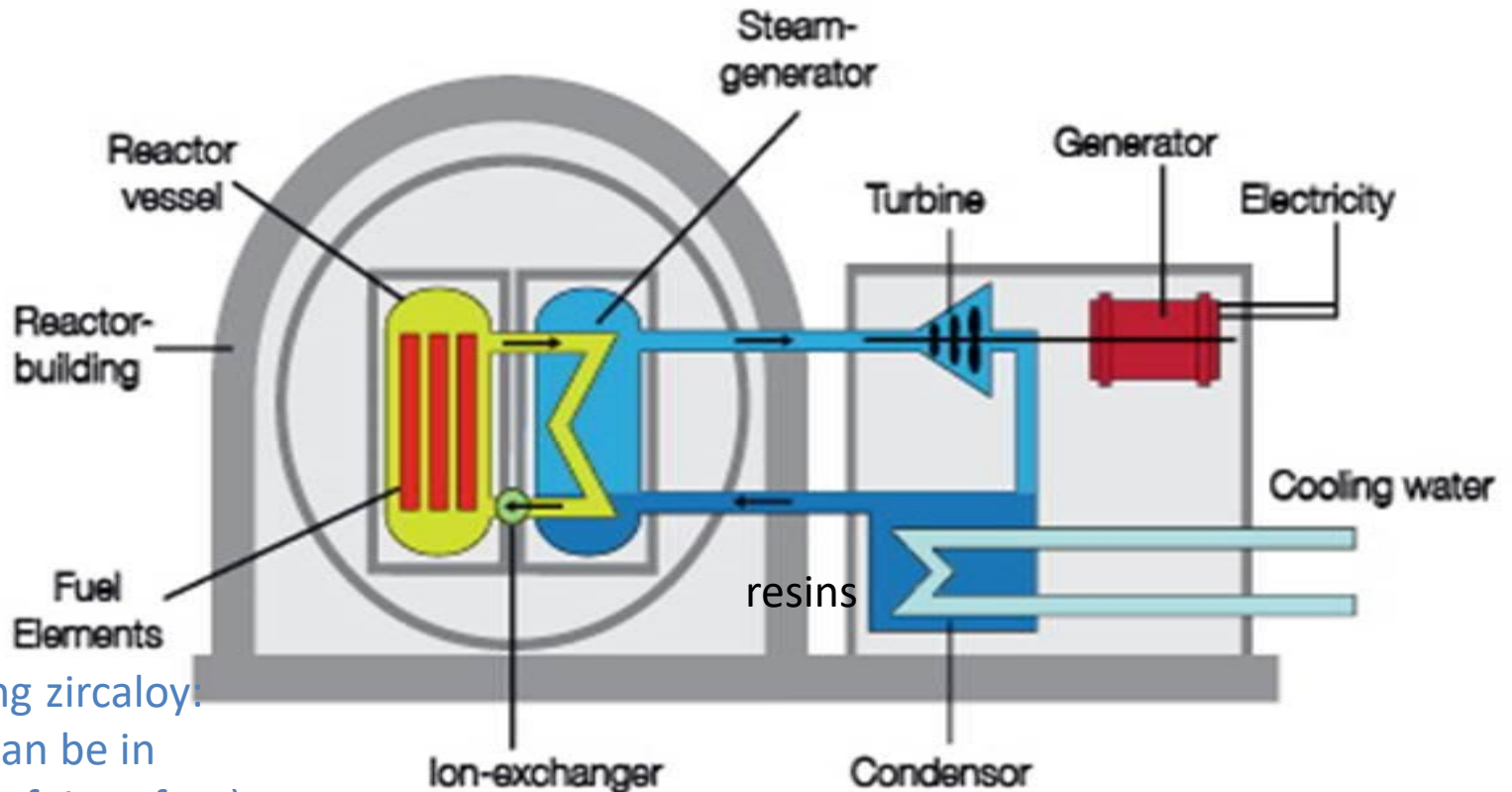


- Knowledge nitrogen impurities for many types of waste
 - EU study (1984) limit nitrogen impurities
 - IAEA (2004) for example
 - limit air ingress primary coolant
 - pH control primary coolant LiOH instead of hydrazine $\text{NH}_2\text{-NH}_2$
- Neutron thermal flux and irradiation period

Origin nitrogen

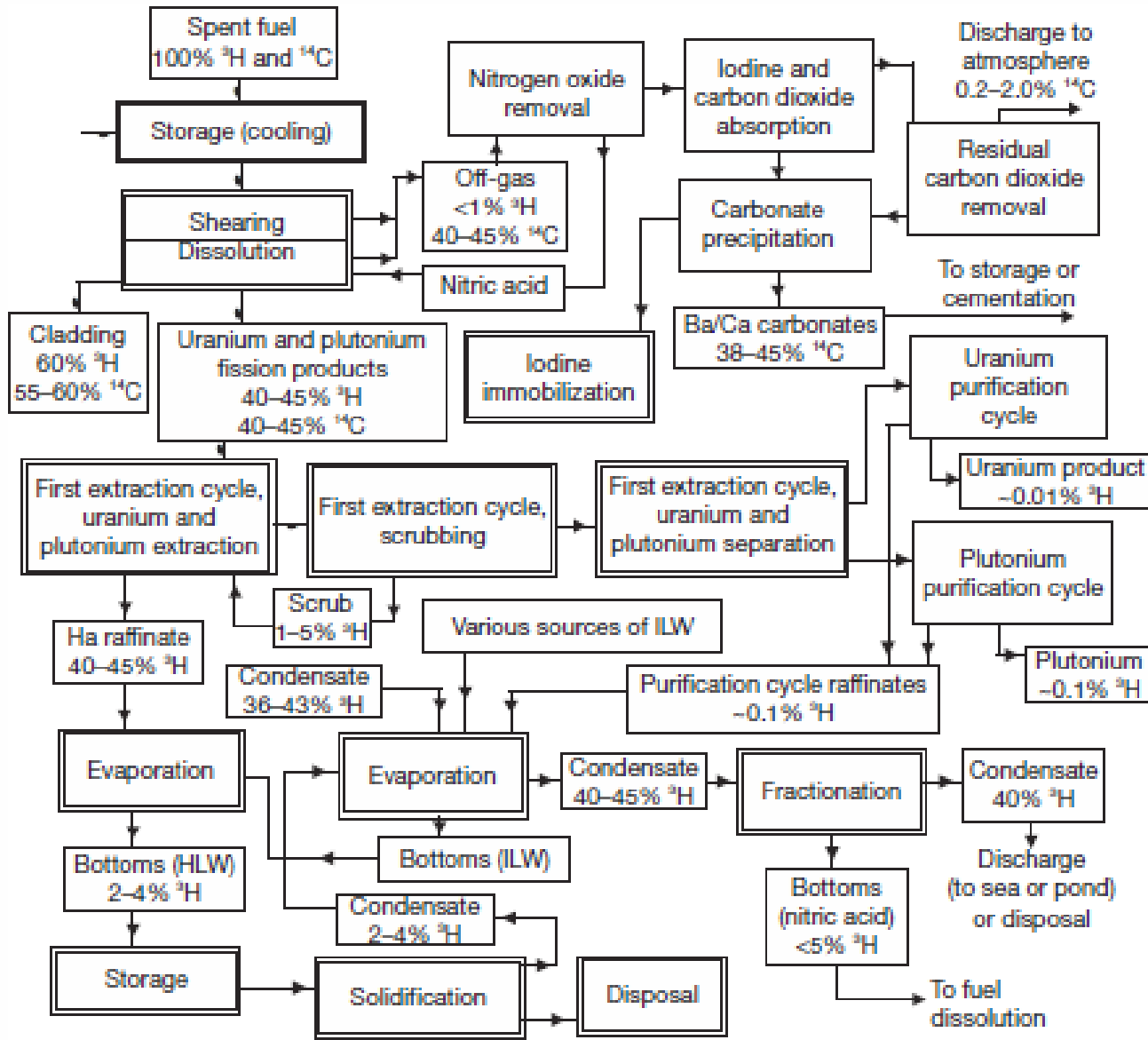
Manufacturing steel: nitrogen can be in
 pig iron,
 Cokes
 Stirring gas

Nitrogen content frequently not reported

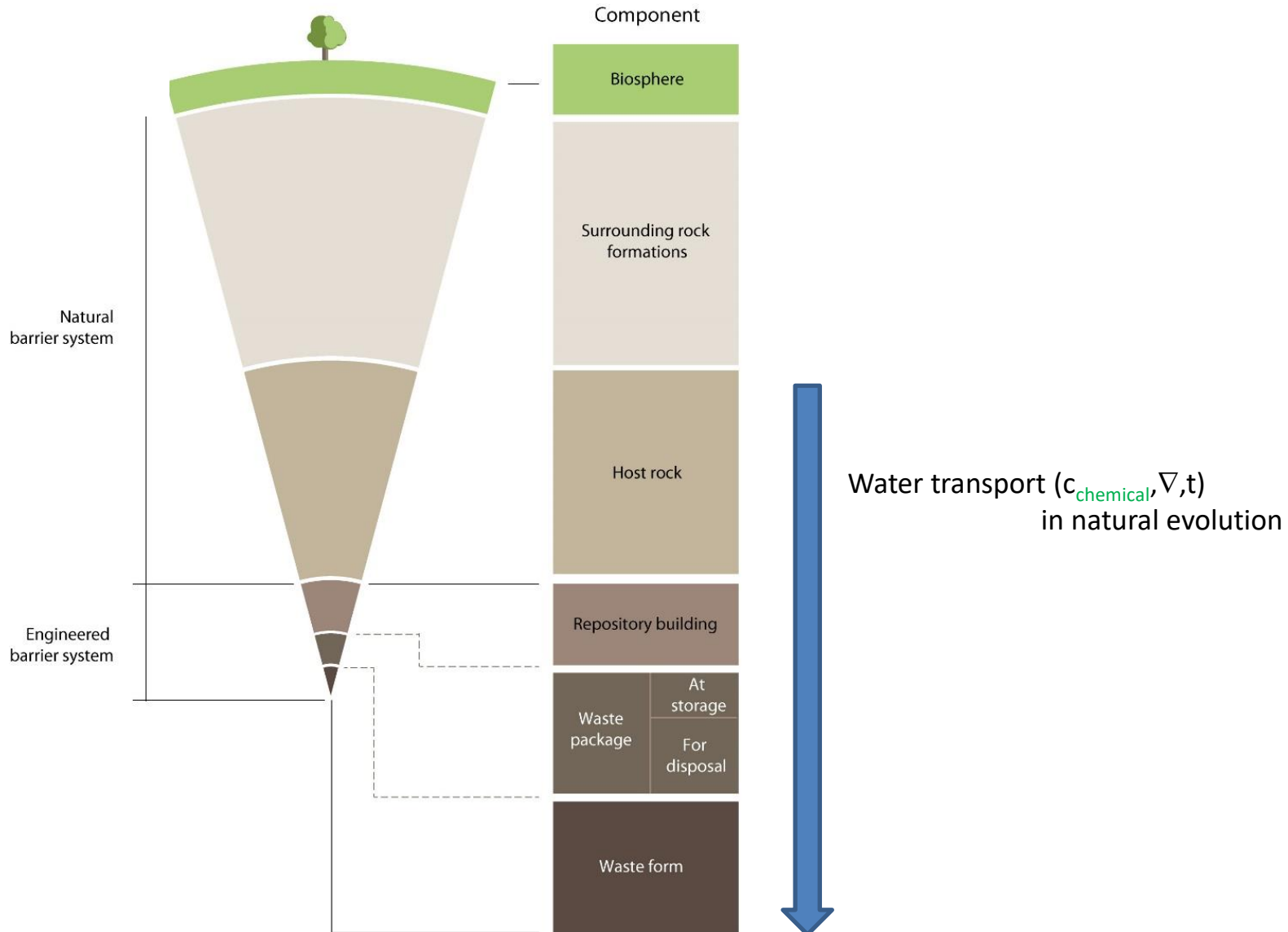


Manufacturing zircaloy:
 nitrogen can be in
 Sponge ingot (Hafnium-free),
 melt

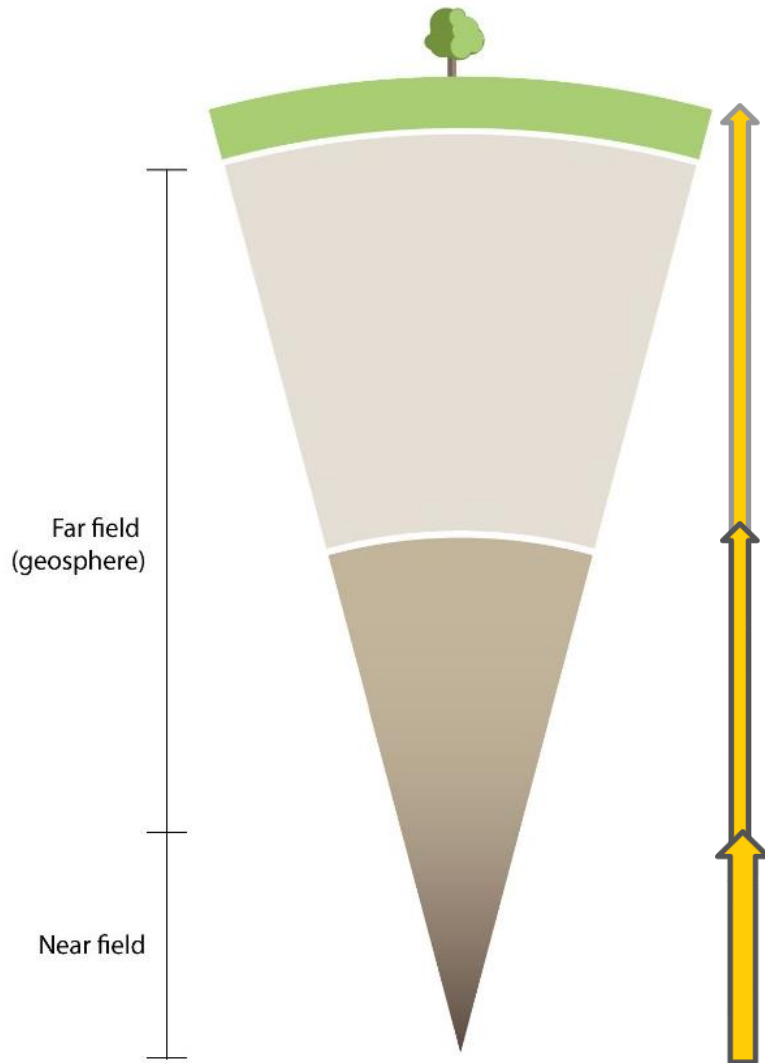
Reprocessing



Geological disposal of waste



Geological disposal of waste



Transport ($c_{\text{chemical} + \text{radionuclides}}, \nabla, t$) in natural evolution
 Dissolved, gas
 for example ^{14}C if CH_4 must be assumed

Transport ($c_{\text{chemical} + \text{radionuclides}}, \nabla, t$) in natural evolution
 Dissolved, ionic
 for example ^{129}I and ^{36}Cl and ^{14}C if HCO_3^- may be assumed

Transport ($c_{\text{chemical} + \text{radionuclides}}, \nabla, t$) in natural evolution
 Retarded by sorption and ultrafiltration
 for example complexes of actinides



Gas, dissolved

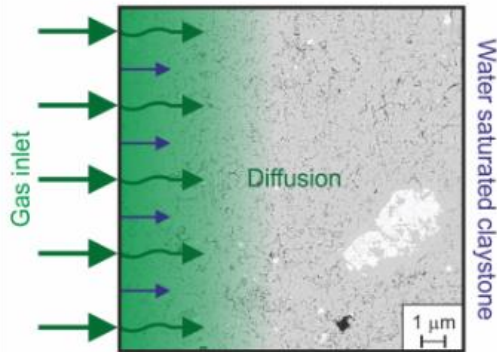


- Waste types investigated in CAST
 - Neutron irradiated metallic compounds
 - Degradation: anaerobic corrosion
 - Hydrogen generation rate
 - Non-metallic neutron irradiated compounds

Free gas, dissolved gas

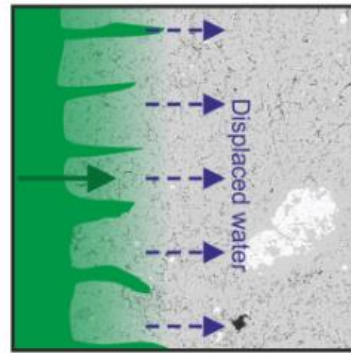
Identified in concrete?

yes

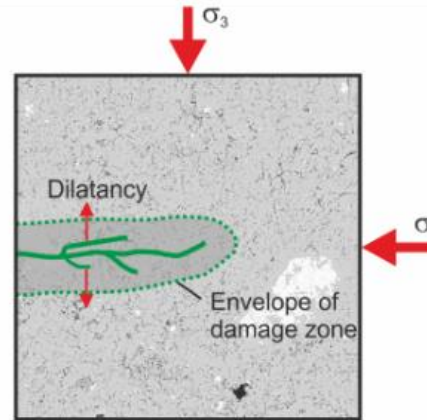


Advection and diffusion of dissolved gas

yes

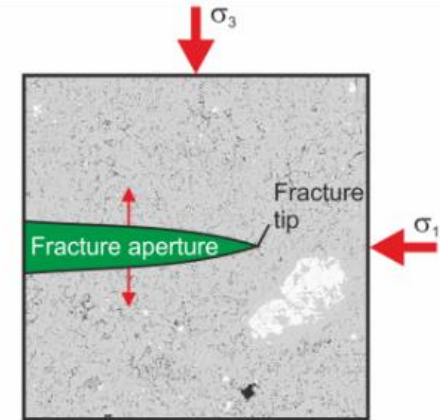


Visco-capillary flow of gas and water phase ("two-phase flow")



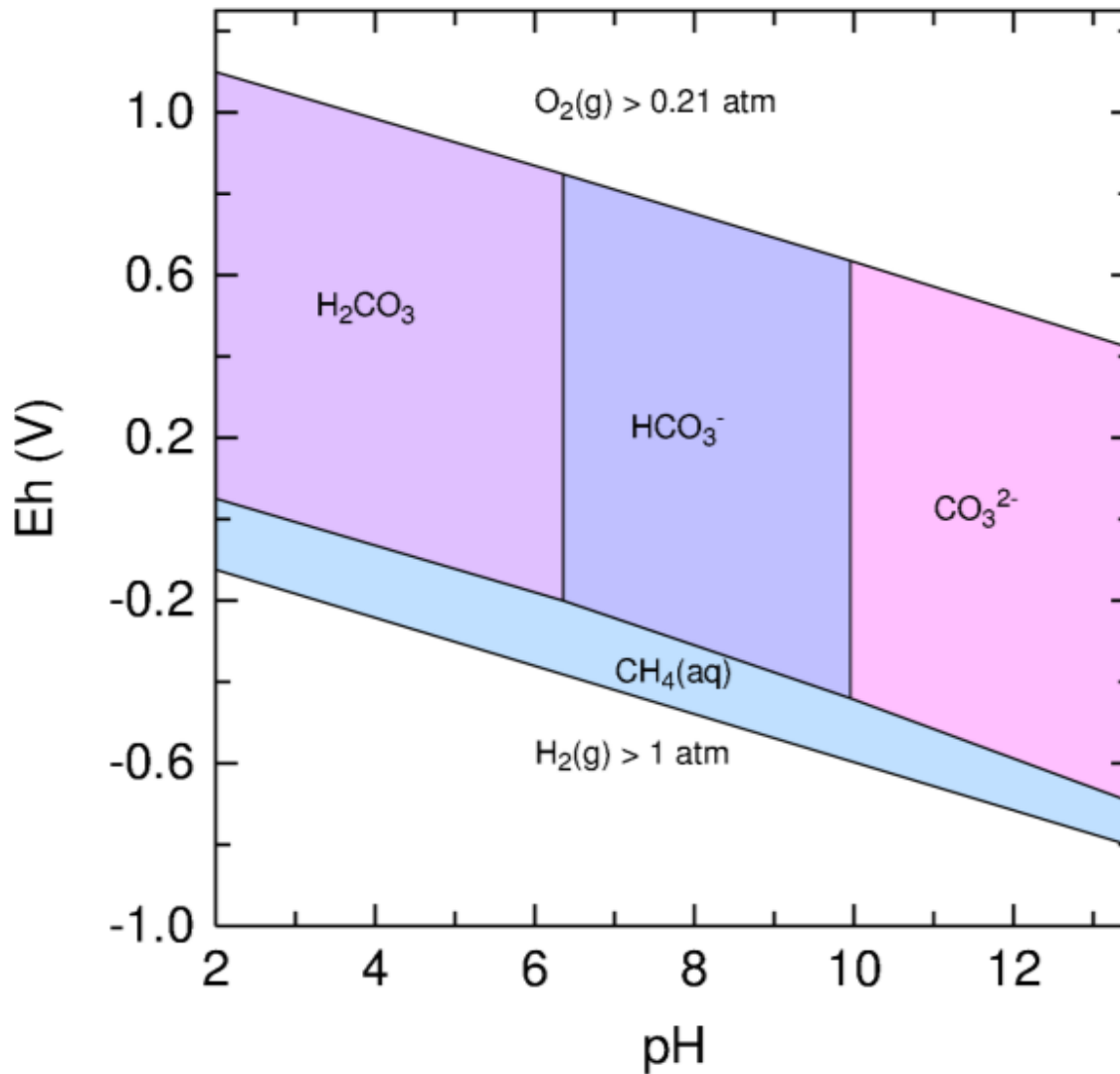
Dilatancy controlled gas flow ("pathway dilation")

yes

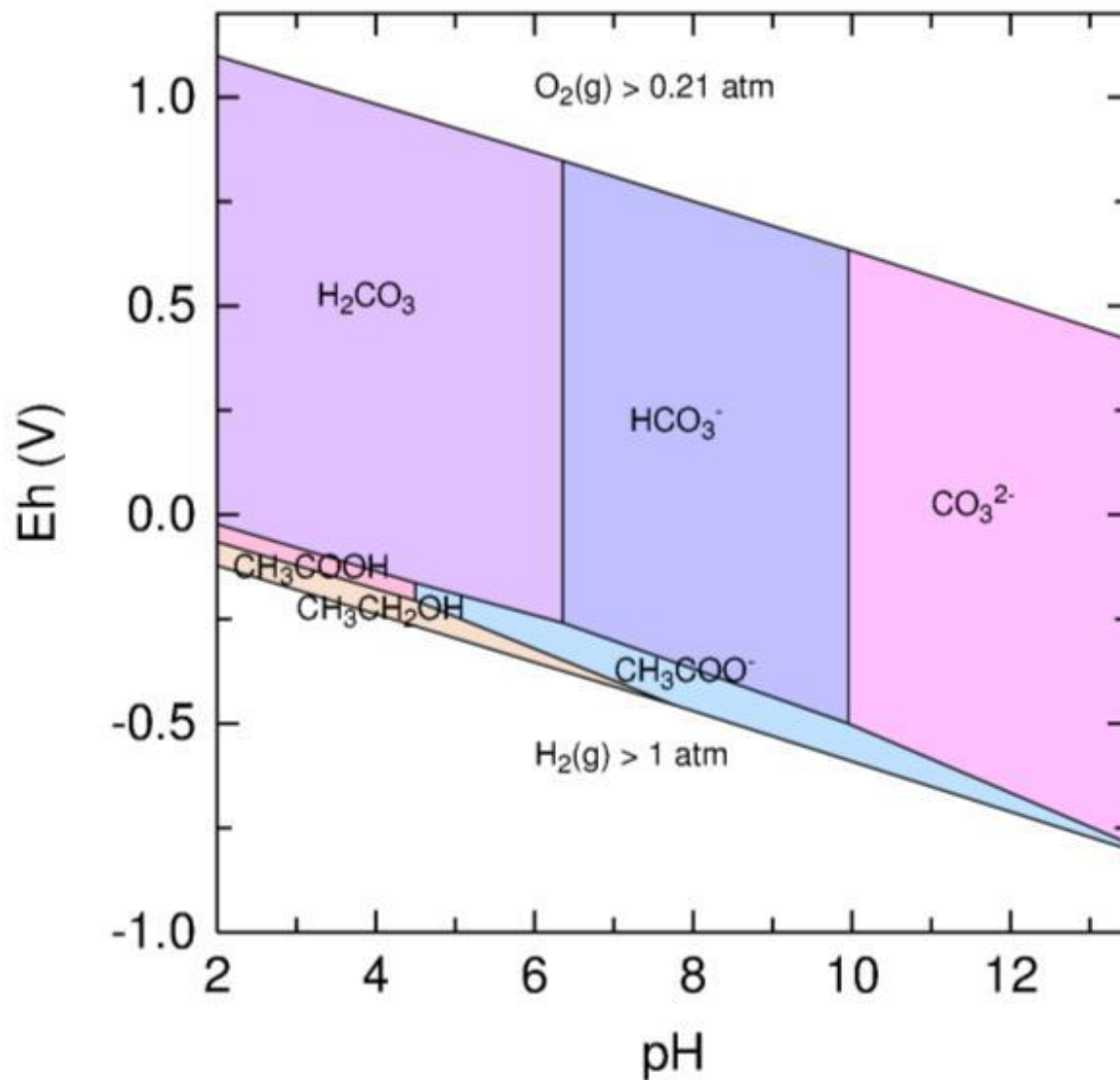


Gas transport in tensile fractures ("hydro-/gasfrac")

Gas, dissolved



Gas, dissolved



Gas, dissolved

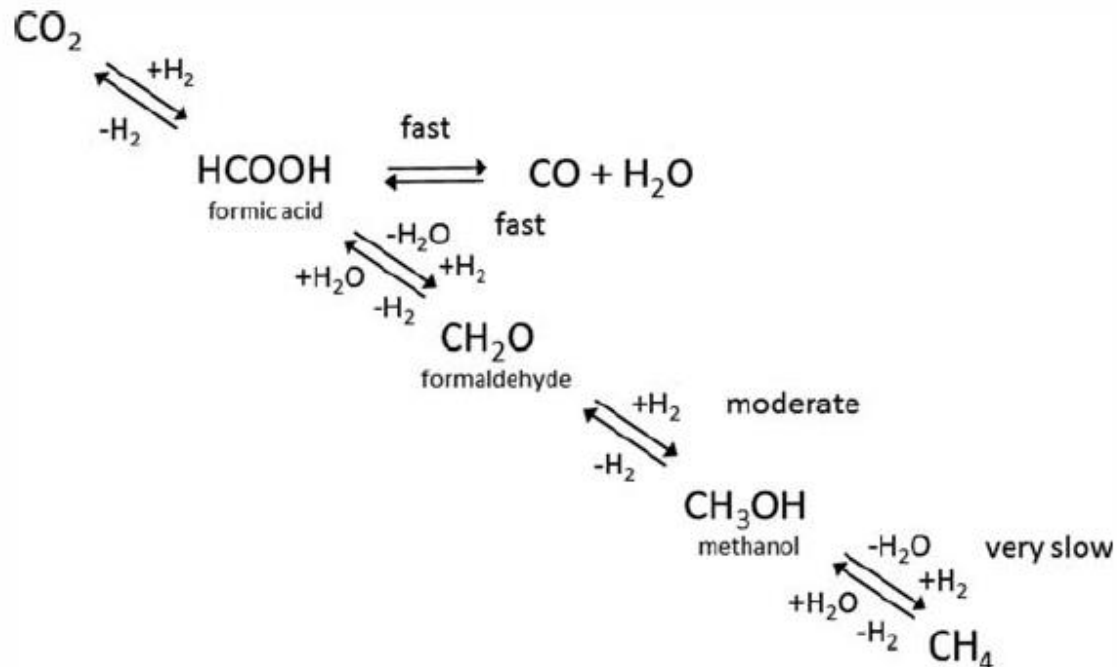
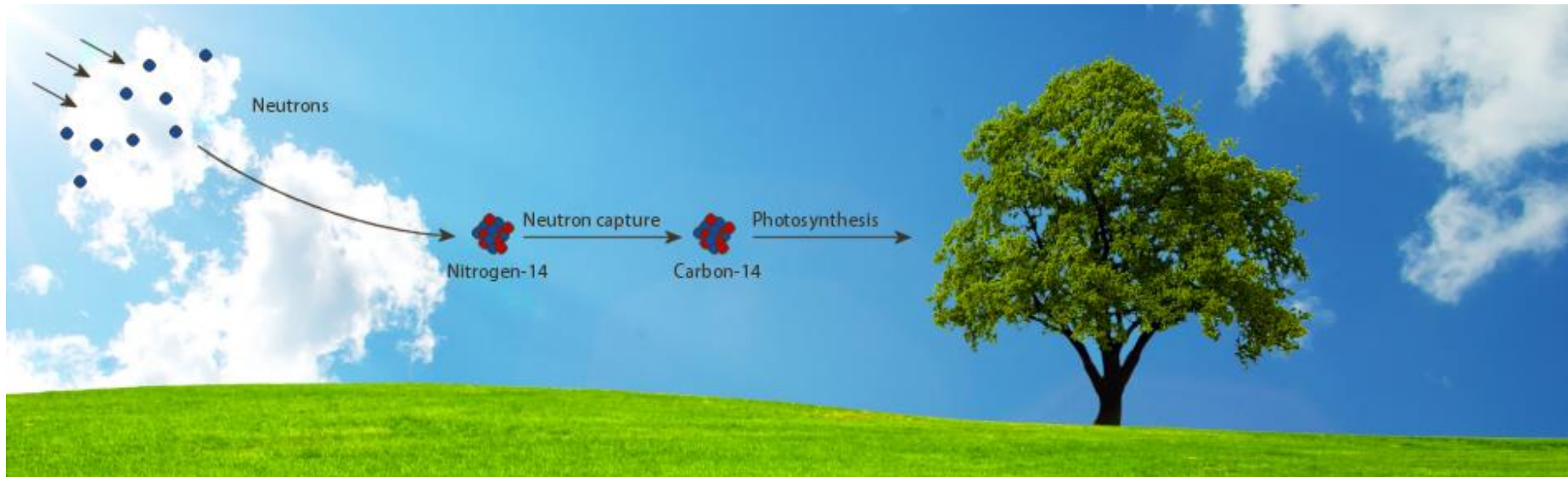


FIG. 4. Reaction scheme for the sequential reduction of CO₂ to methane (modified from McCollom and Seewald, 2007).

Natural carbon-14





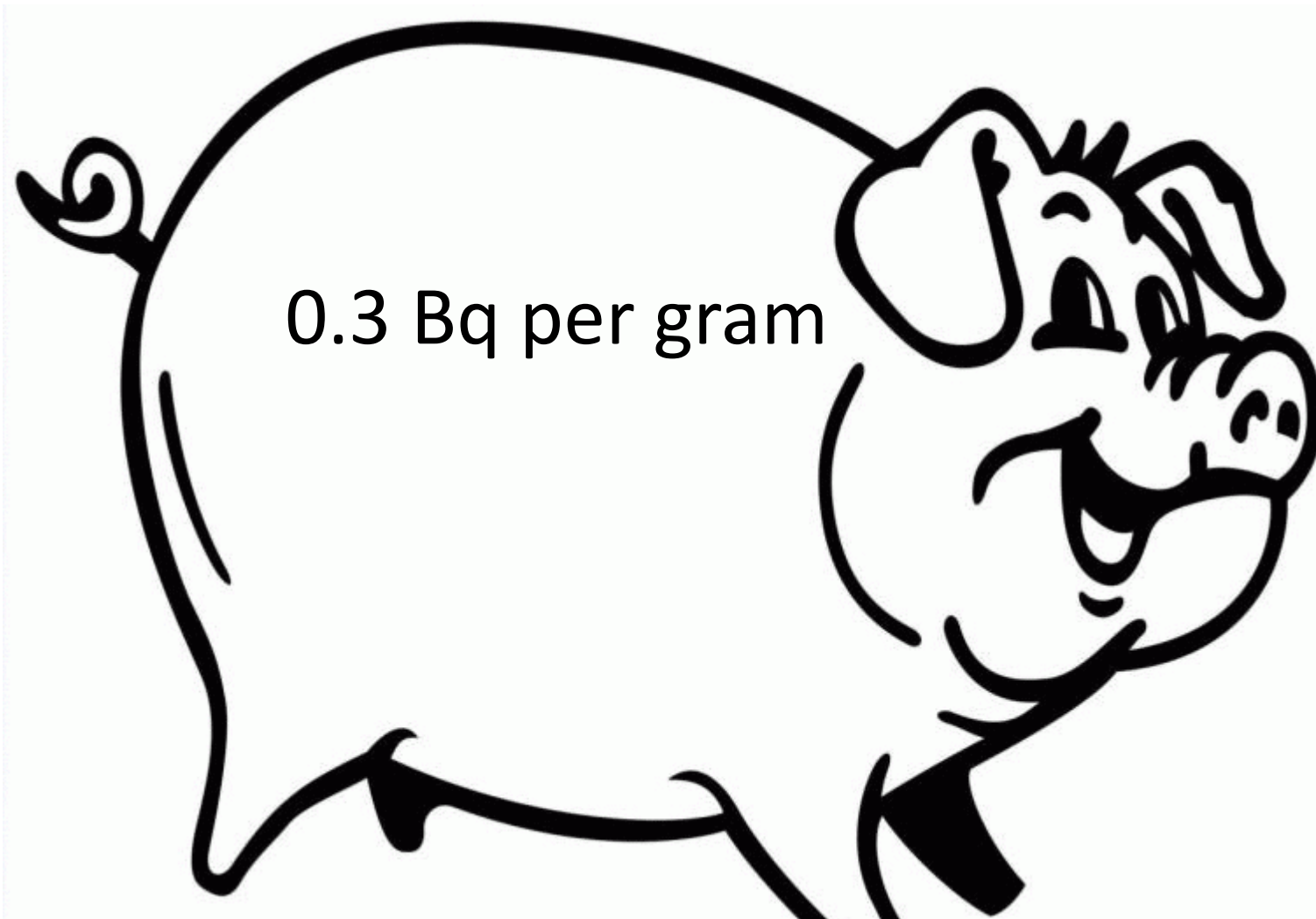
Exposure paths



- Inhalation
 - No concentration factor if not used by living matter for example noble gases
- Radiation exposure
 - For DTM radionuclides not likely
- Ingestion
 - Concentration factor if taken up by living matter for example carbon
 - Accumulation ^{14}C

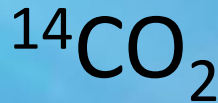


Natural carbon-14

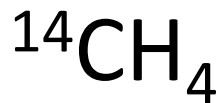


0.3 Bq per gram

Main exposure path for humans: ingestion IRSN,2010

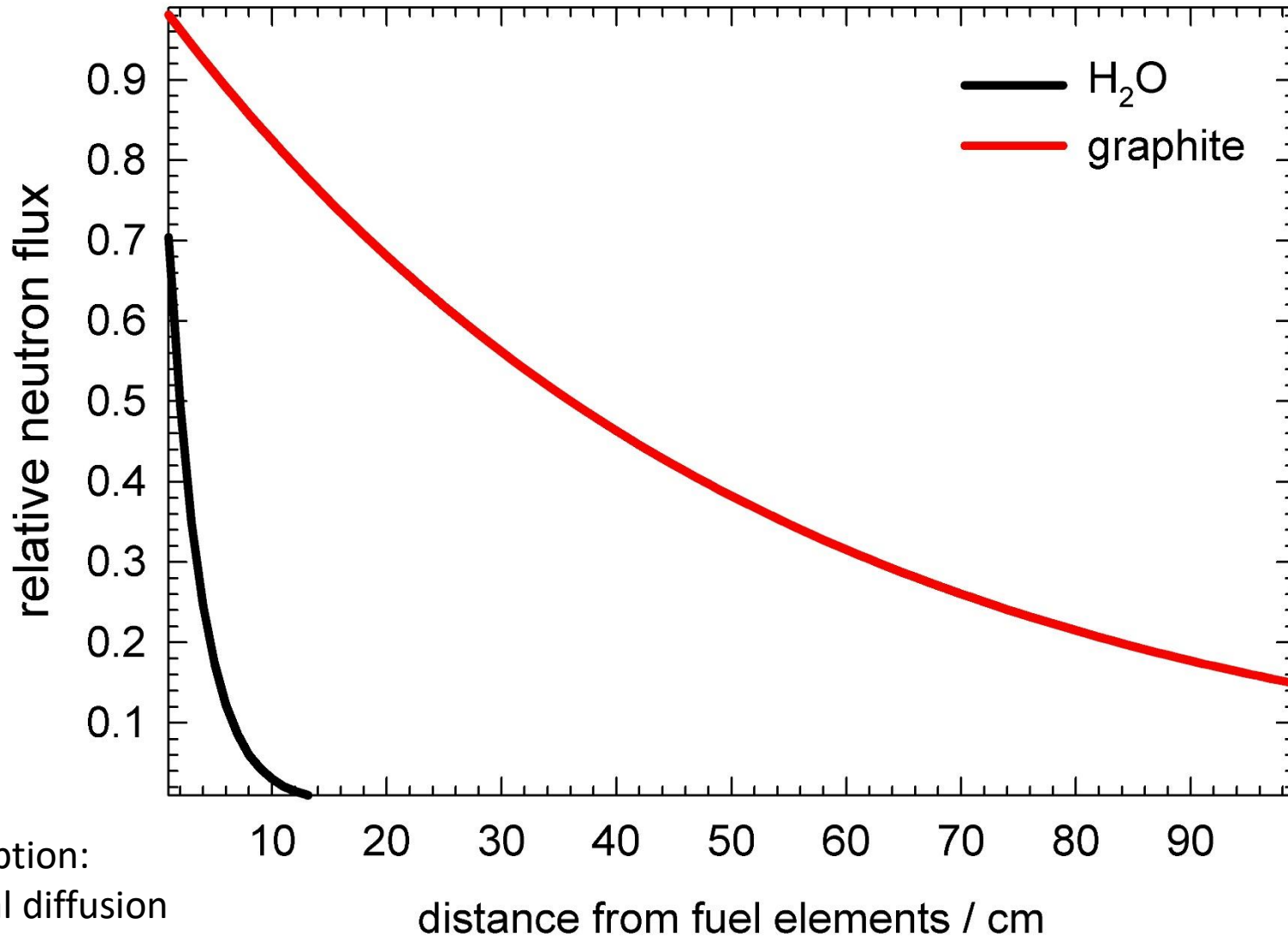


root zone, microbial oxidation

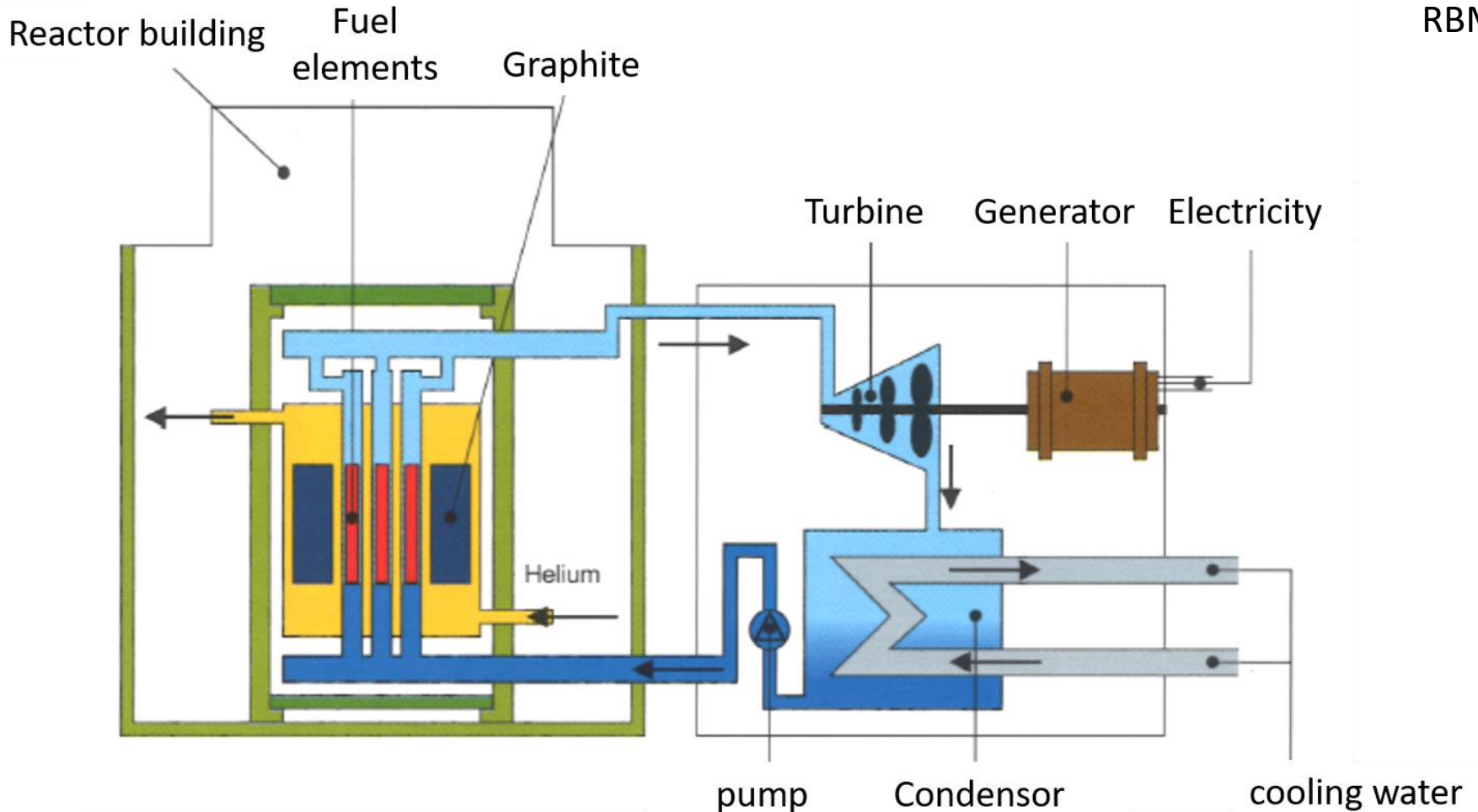


Main exposure path: ingestion

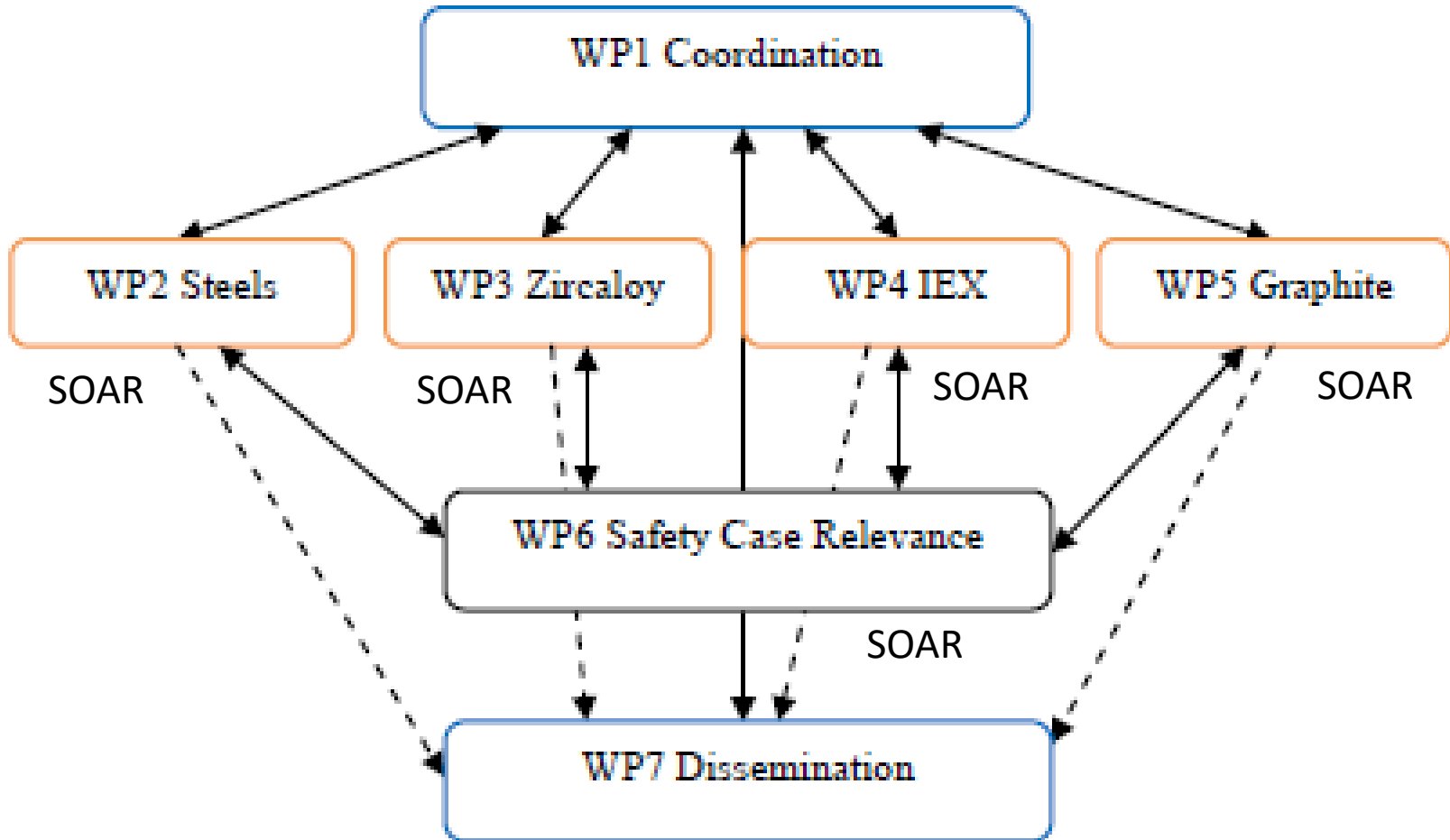
Generation

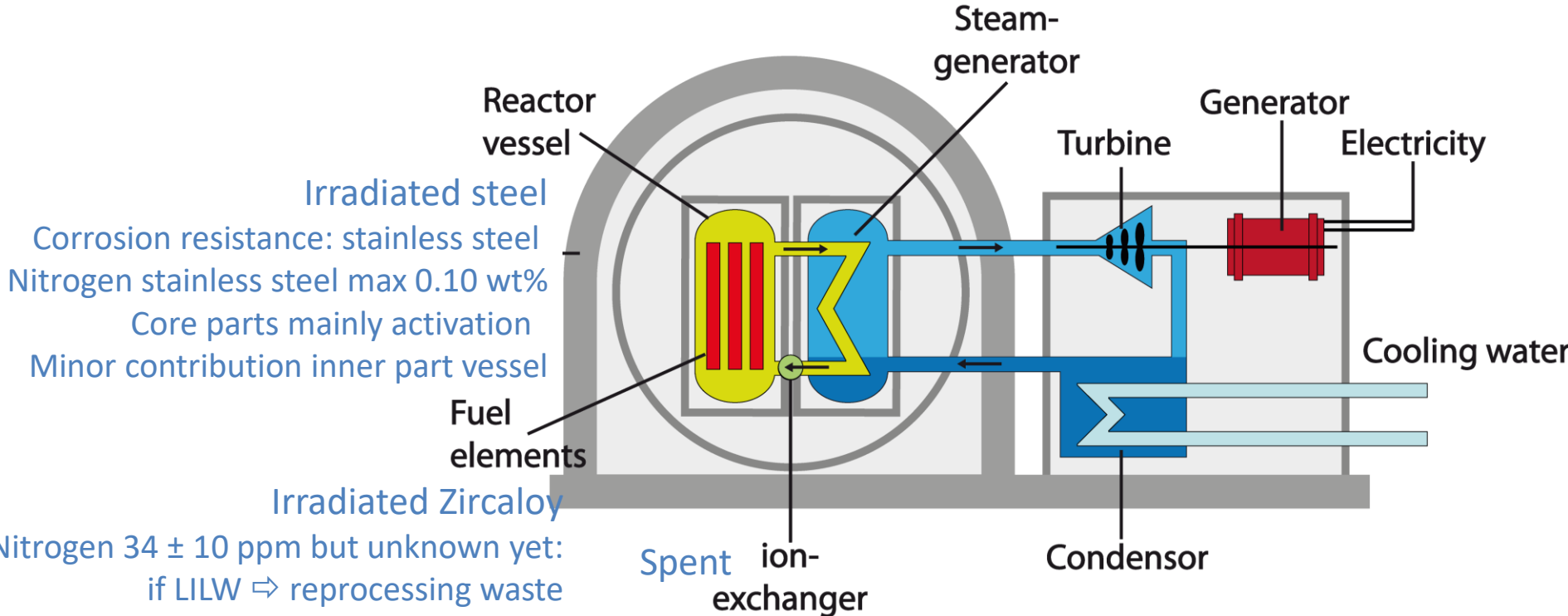


Assumption:
thermal diffusion



Irradiated graphite, in CAST mainly LILW: moderator graphite instead of water
 EU research project CarboWaste because HLW engineered barrier →
 expected containment period several decay times of carbon-14 ($t_{1/2}=5730$ years) Toulhout,2015





Irradiated steel
 Corrosion resistance: stainless steel
 Nitrogen stainless steel max 0.10 wt%
 Core parts mainly activation
 Minor contribution inner part vessel

Irradiated Zircaloy
 Nitrogen 34 ± 10 ppm but unknown yet:
 if LILW \Rightarrow reprocessing waste
 1) mixture with (foreign) wastes
 2) Inconel ends
 3) additional contribution capture gaseous carbon-14

Unknown yet:
 oxygen-17 reduction in 10^3 cross section included

Irradiated graphite (in CAST mainly LILW):
 if nitrogen content larger than 15 ppm \Rightarrow main contribution to carbon-14

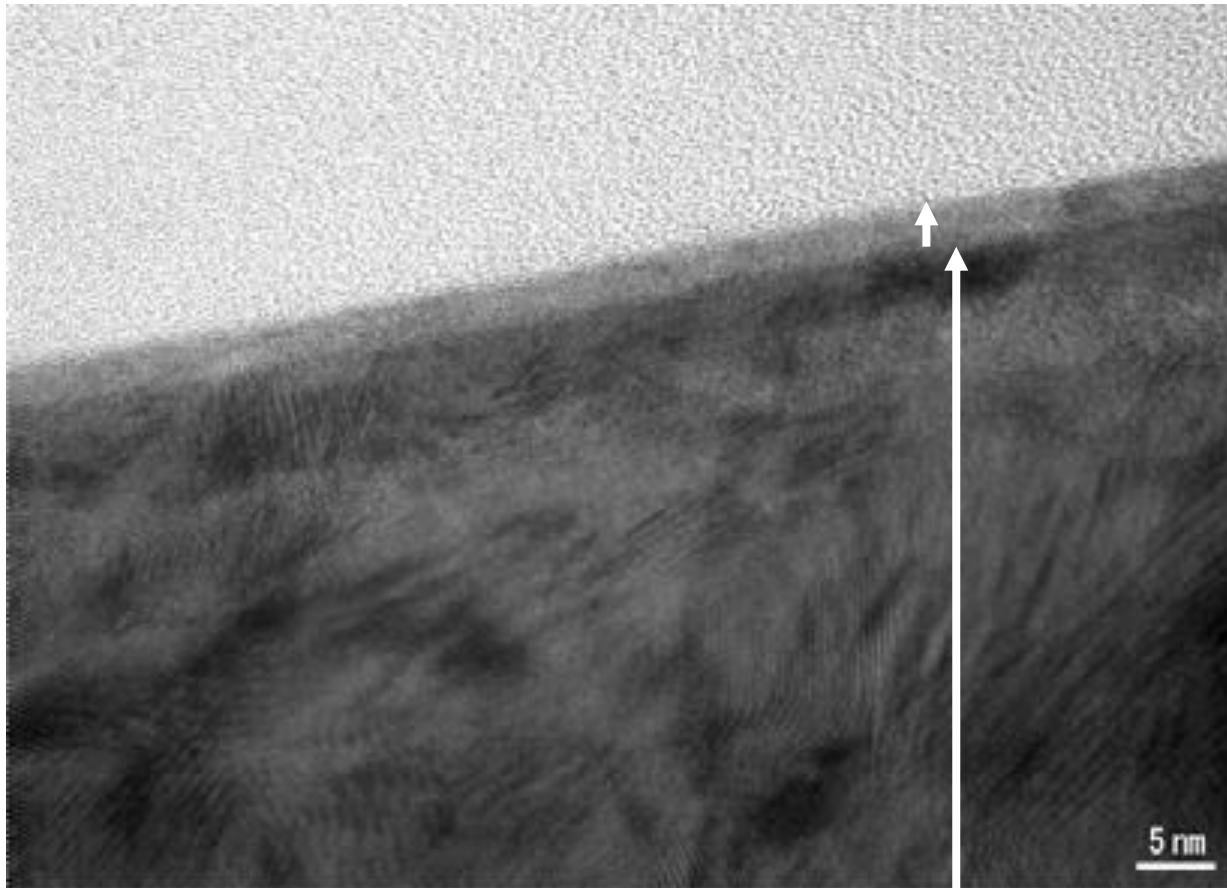


Neutron irradiated steel



- Core assumed 10^5 Bq per gram
- Outer parts for example vessel assumed 10^3 Bq per gram
 - Sample vessel available in CAST 18 Bq per gram steel

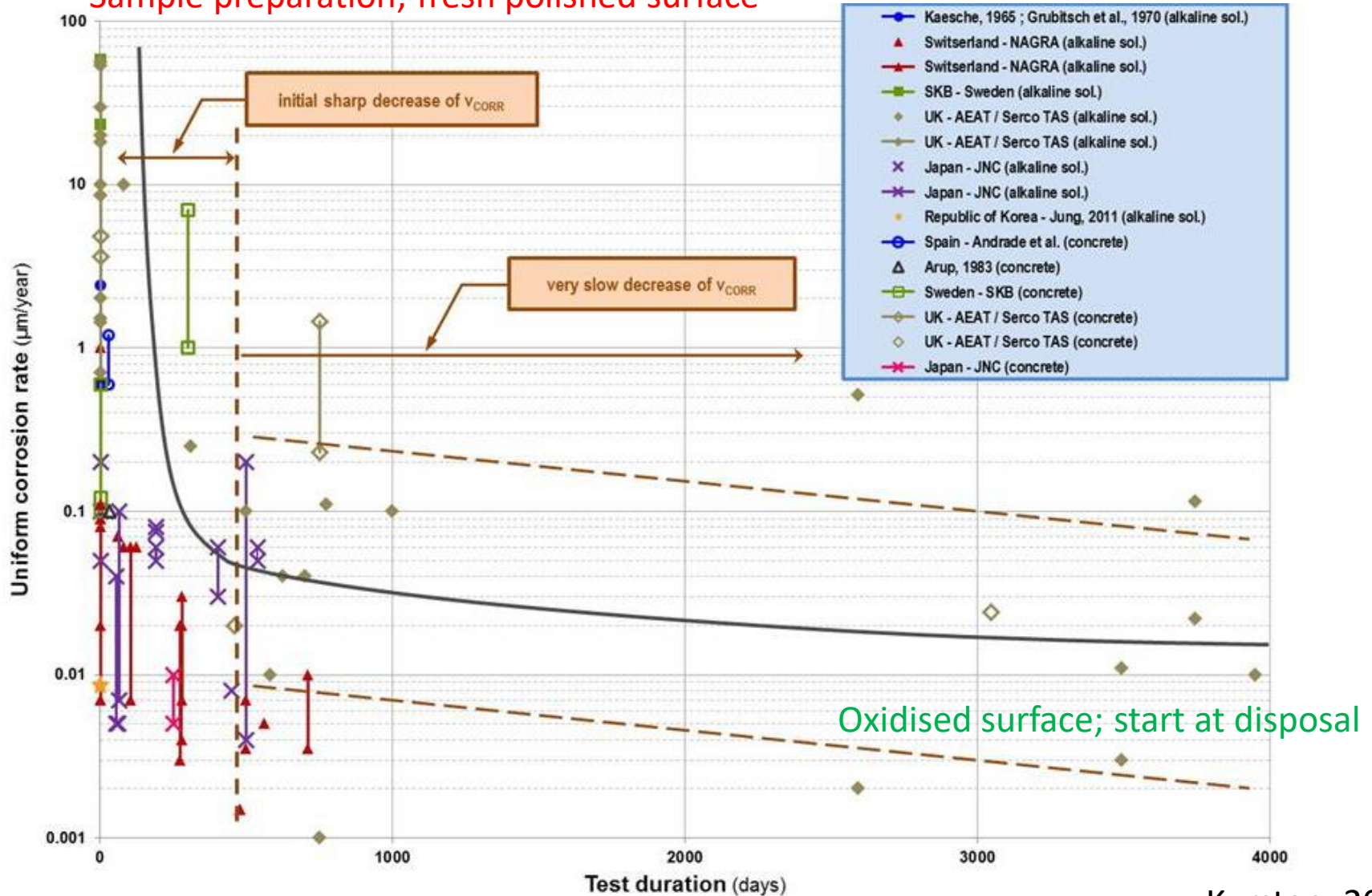
Neutron irradiated steel



Steel



Sample preparation; fresh polished surface



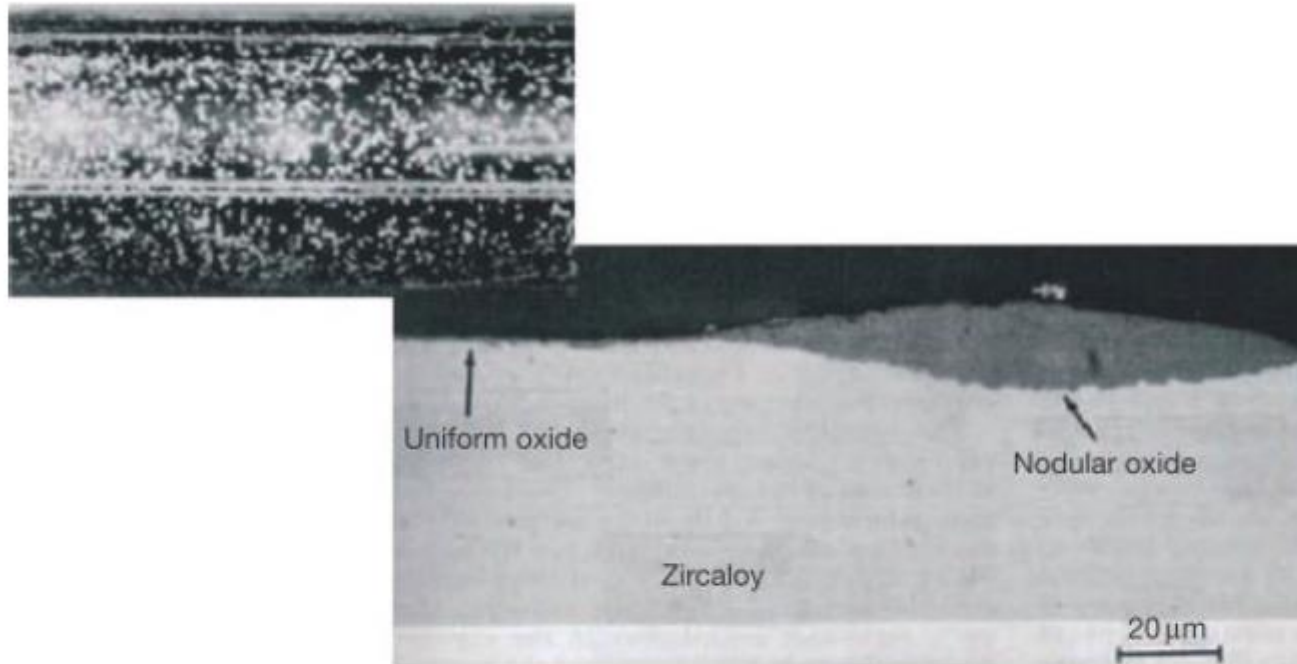


Neutron irradiated Zircaloy

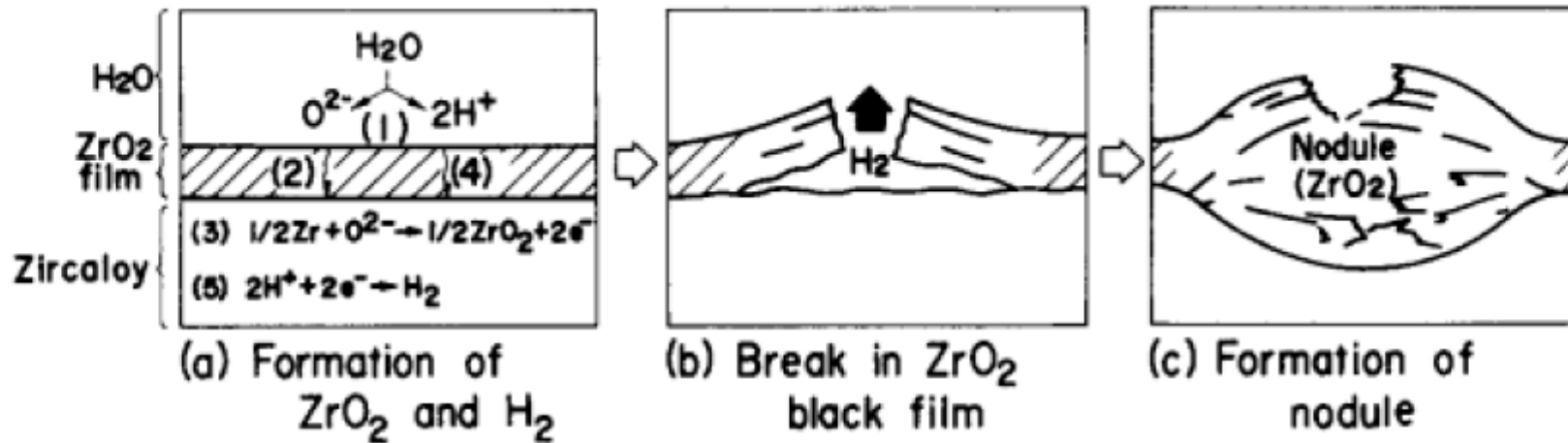


- $\approx 10^4$ Bq ^{14}C per gram Zircaloy
 - Tenfold lower nitrogen content than steel
 - Operational waste not decommissioning waste consequently smaller neutron irradiation period
- Carbon solubility smaller than nitrogen solubility
 - Small precipitate 14-carbides

Neutron irradiated Zircaloy

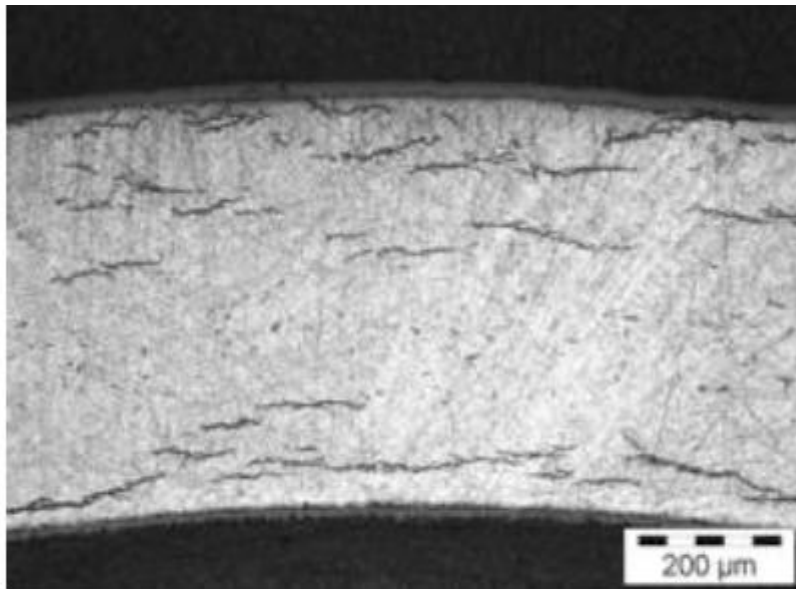


Neutron irradiated Zircaloy

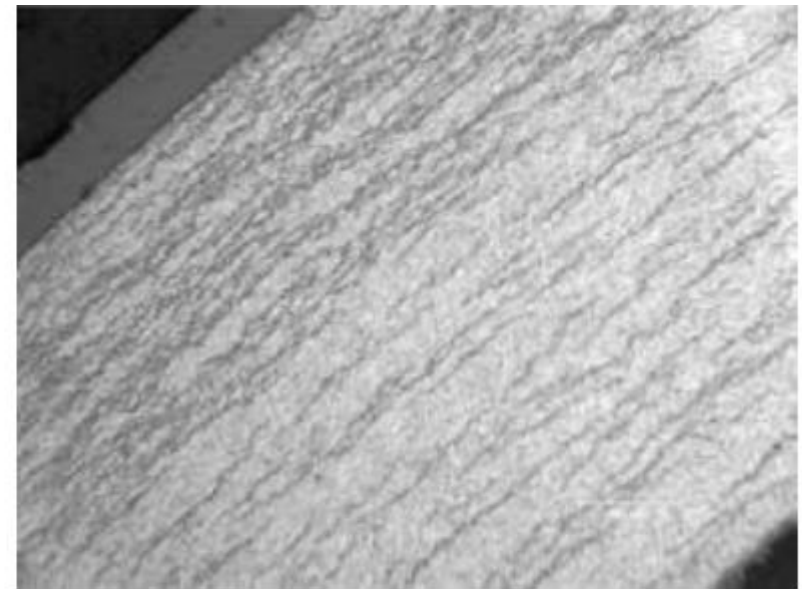


: Typical appearance of nodular corrosion in visual inspection and metallographic examination [ALL 2012] and mechanism of formation of lenticular nodules [KUW 1983]

Neutron irradiated Zircaloy



M5™, 5 cycles, stage 6
[AMB 2010]



Zirlo™, 67 GWd.t⁻¹
[AND 2012]



CSD-c as stored at COVRA's storage facility;

typical value for 900 MW NPP

1.4×10^{10} ^{14}C Bq per container

27000 Bq / gram solid waste

528 kg: 393 kg Zr (hulls) , 19 kg Inconel (ends) , 116 kg ss (technological waste)

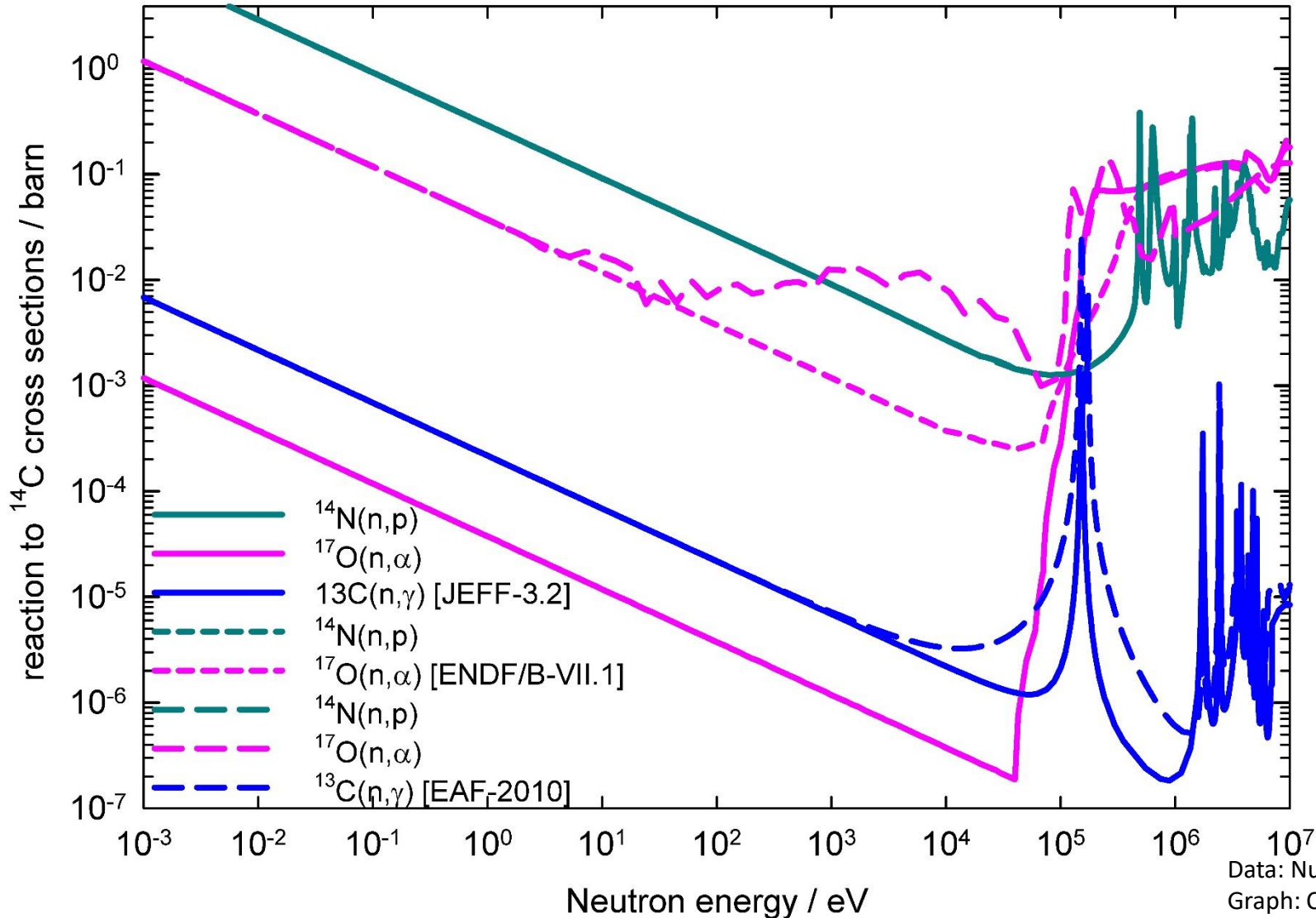
Reported by AREVA

GK280
07/R115

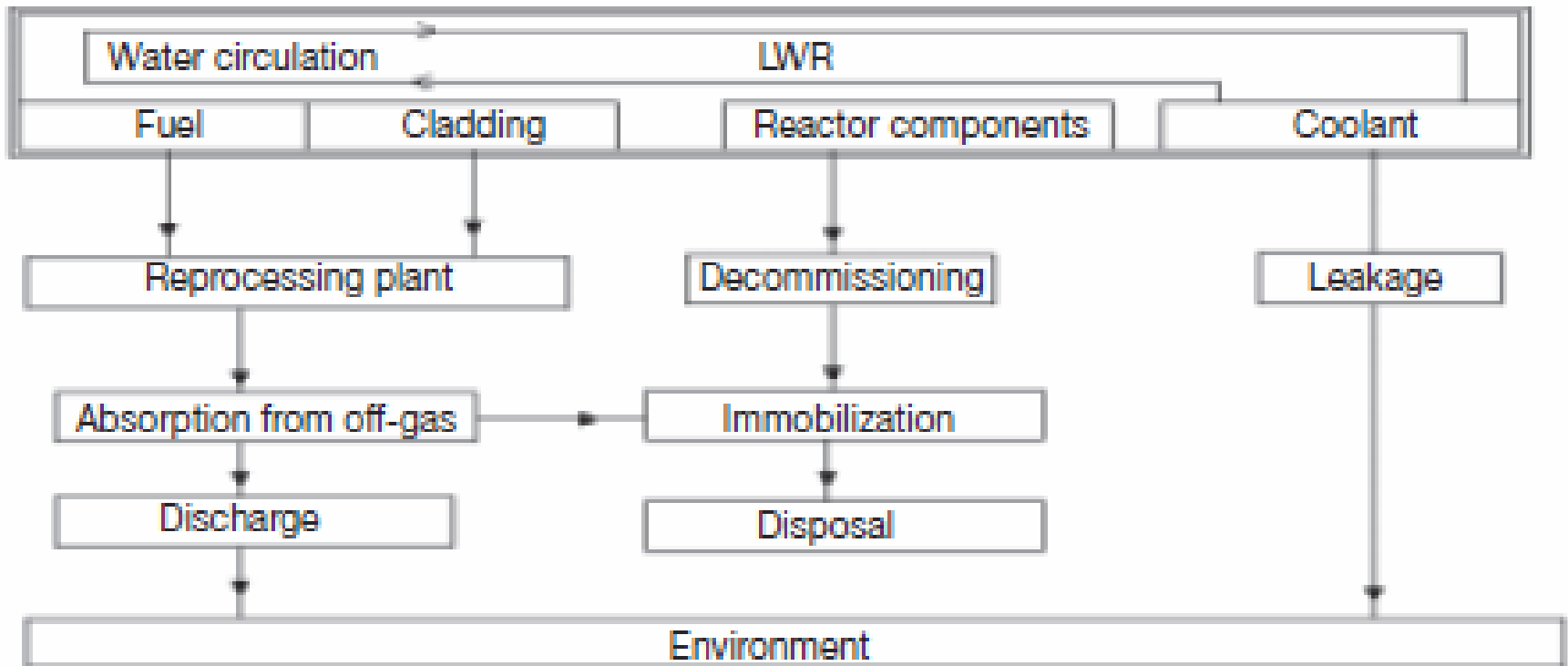
Spent ion exchange resins



at 300 K



Spent ion exchange resins



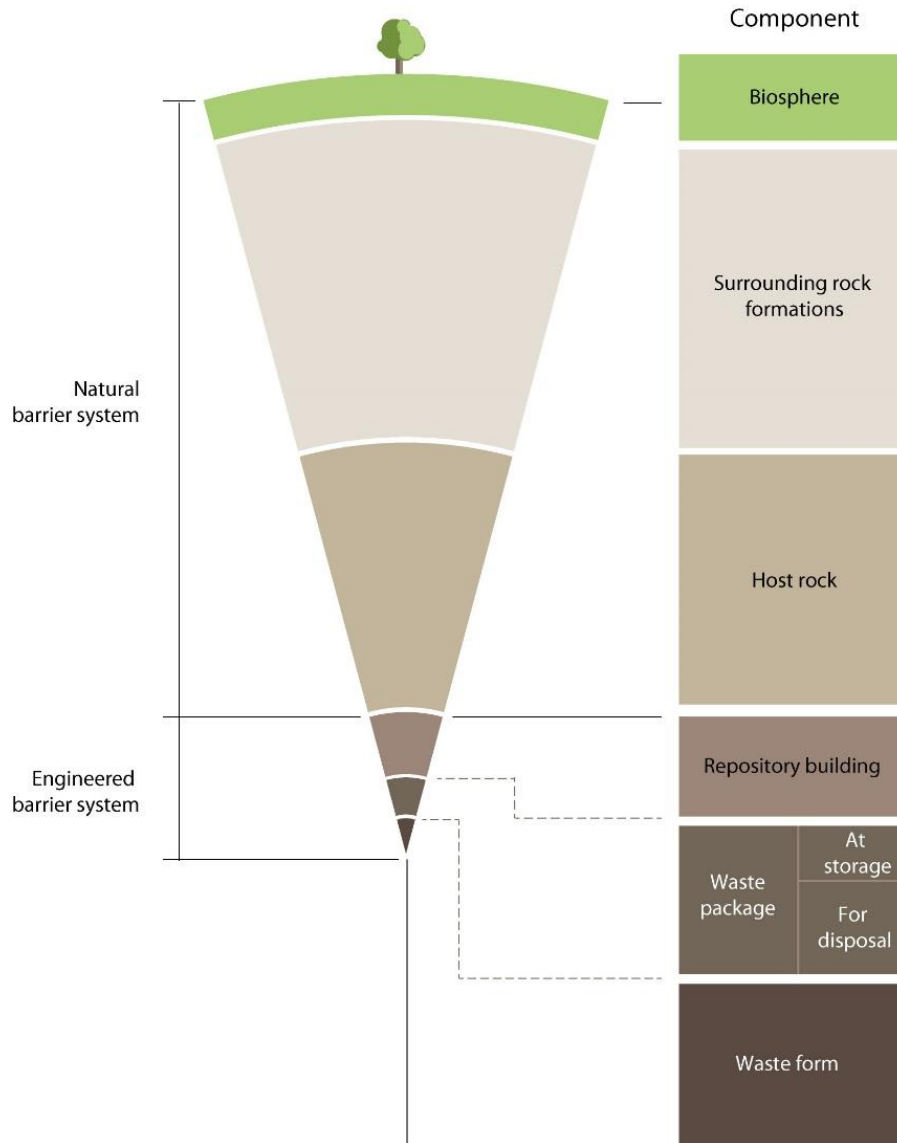
Neutron irradiated graphite

- Romania: contact-handled irradiated graphite
 - see first CAST Newsletter



- Italy: remote-handled irradiated graphite
 - Canzone G et al (SOGIN) Dismantling of the graphite pile of Latina NPP: Characterization and handling/removal equipment for single brick or multi-bricks, Progress in Nuclear Energy 93 (2016) 146-154

Release mechanism



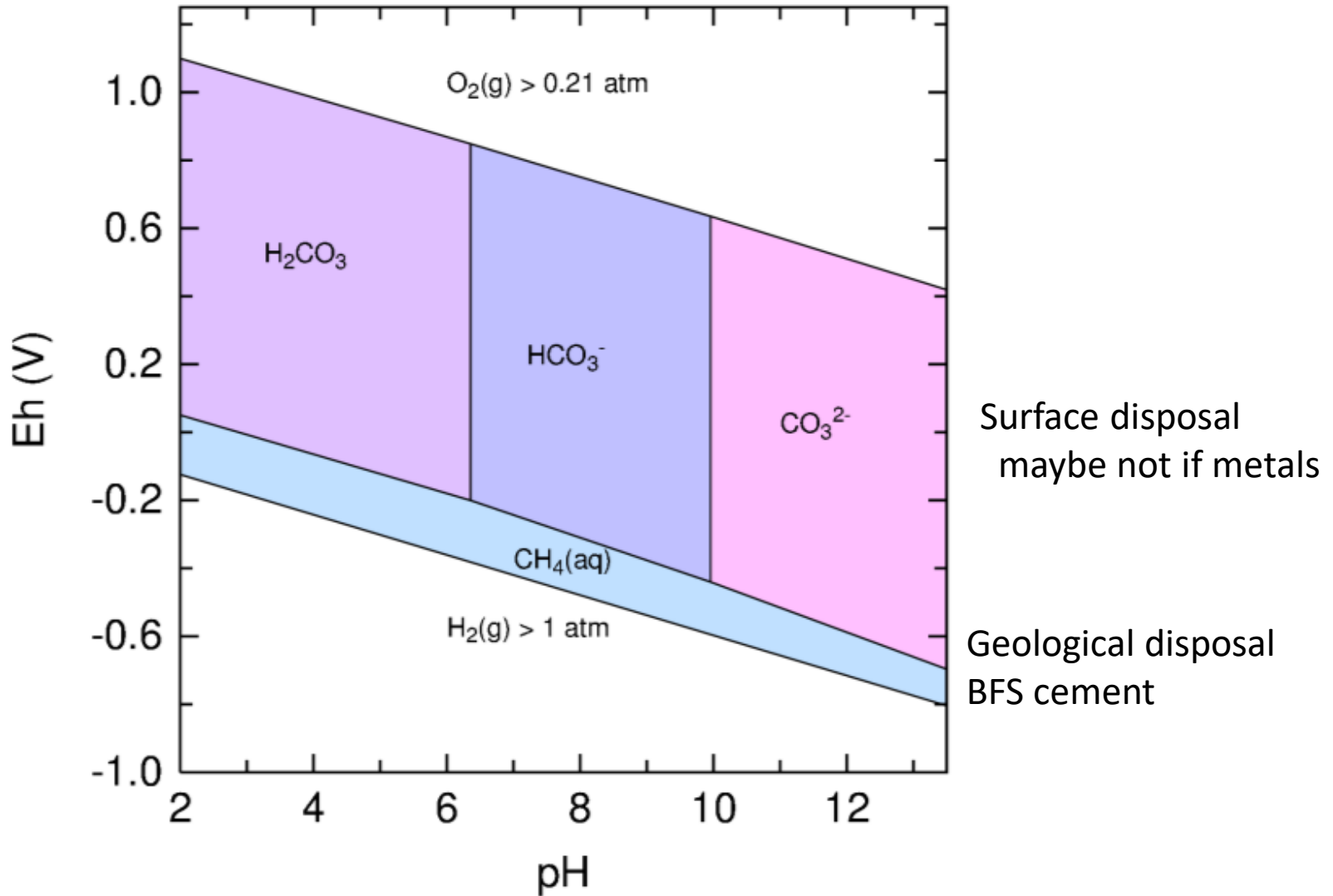
Cementitious materials



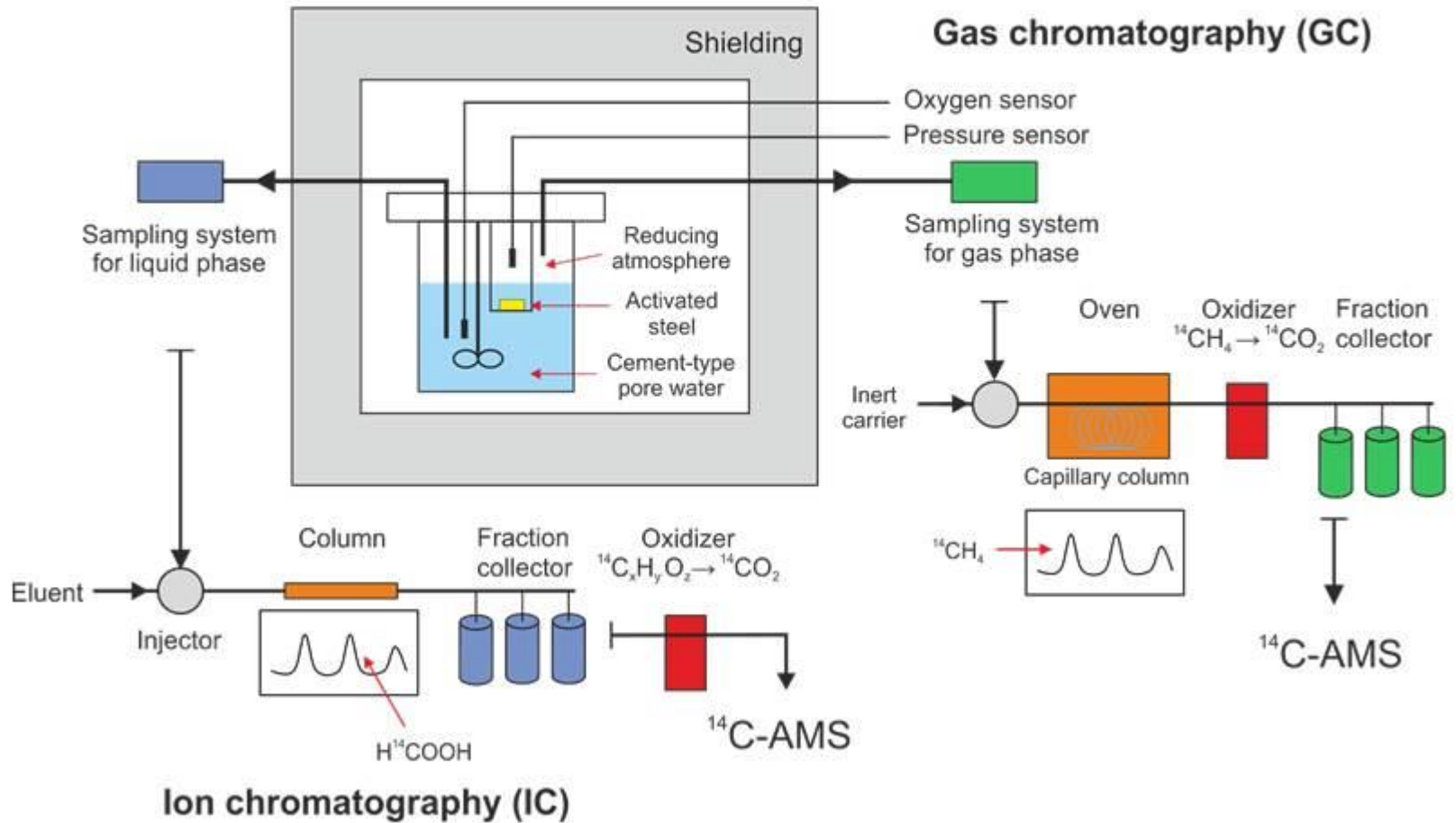
Release mechanism

- Source term: carbon-14 release rate or rates from waste
- Release under conditions relevant for waste packaging and disposal to underground facilities
 - Cementitious matrices, main waste packaging conditions considered in CAST
 - Cement alkaline conditions
 - Portland: initially slightly oxidising and largely unbuffered because of lack of electroactive species
 - corrosion of metals may reduce redox potential locally
 - Blast furnace slag: initially reducing due small amount of FeS_2 – blueish colour when not oxidised –
 - corrosion of metals may locally sustain reducing conditions
 - Underground
 - Near-surface disposal: aerobic exposure conditions
 - Deep geological disposal: anaerobic exposure conditions

Speciation



Speciation measurement





Conclusions / highlights



- CAST finishes on 1 April 2018
 - Final General Assembly Meeting in January 2018 in France (Lyon)
 - Submission abstracts before 1 September 2017
- During running research programme CAST
 - State Of the Art reports at start of the programme
 - knowledge management
 - End-User view: what does experimental research contribute to what is already known?
 - Determination activity concentration of carbon-14 in waste appeared to become more important
 - Nitrogen impurities in steel measured
 - Unknown if nitrogen content has been reduced since 1984 ALARA
 - Main presence carbon-14 in processed waste, in Zircaloy hulls, not yet known
 - Focus on reliable determination of carbon-14 activity concentration in spent ion exchange resins, speciation of ionic carbon not yet identified
 - Also in neutron irradiated graphite, nitrogen impurities can be main source of carbon-14
 - Obtaining representative samples and setting-up experiments takes time
 - Corrosion rates of steel at geological disposal, i.e. passivated surfaces in cementitious materials, perhaps too hard to measure reliably DTM radionuclide carbon-14 release rate
 - Nitrogen at impurity level expected to be dissolved in metal lattice consequently congruent release of carbon-14 may be expected if migration of carbon within steel at reactor conditions can be neglected



Plan of content training course



- Like PETRUS-ANNETTE free of charge
 - Have a look at www.projectcast.eu/training/courses for further details
- Plan of content
 - For the types of waste investigated in CAST
 - Generation of carbon-14 in nuclear plant
 - Calculate potential release of radionuclides
 - Scheduled at the end of 2017 / early 2018
 - probably in the Netherlands in order to visit the waste investigated in CAST
 - About two days
 - Please send an e-mail to erika.neeft@covra.nl if you are interested preferably after your summer holiday



Thank you for attention
any questions?



References



CAST reports and newsletters free online available at www.projectcast.eu

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